

Science Bowl Questions – Chemistry, Set 2

1. Multiple Choice: The triple bond between the carbon atoms causes acetylene, C_2H_2 , to have which of the following shapes?

w) trigonal planar (pron: try-gon-al)
x) linear
y) tetrahedral
z) trigonal bipyramidal

ANSWER: X -- LINEAR

2. Short Answer: What is the EMPIRICAL formula for glucose?

ANSWER: CH_2O (read: C - H - 2 - O)

3. Multiple Choice: Which of the following four elements has the largest atomic radius? Is it:

w) strontium
x) francium
y) calcium
z) bromine

ANSWER: X -- FRANCIUM

4. Multiple Choice: Which of the following odors does the ester, methyl salicylate, produce? Is the odor that of:

w) cinnamon
x) avocado
y) orange
z) oil of wintergreen

ANSWER: D -- OIL OF WINTERGREEN

5. Short Answer: What two chemical elements are found in sphalerite?

ANSWER: ZINC AND SULFUR (ZnS)

6. Short Answer: Periclase is composed of what two elements?

ANSWER: MAGNESIUM AND OXYGEN (MgO)

7. Short Answer: Who patented the common dry cell?

ANSWER: GEORGES LECLANCHÉ

8. Multiple Choice: $Ca(BrO_3)_2$ (read: C - a, left parenthesis, B - r - O - subscript 3, right parenthesis, subscript 2) is called:

w) calcium bromate
x) calcium bromite
y) calcium dibromite
z) calcium bromide

ANSWER: W -- CALCIUM BROMATE

9. Short Answer: What is the common oxidation state of Radium?

Science Bowl Questions Chemistry With Answer

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Committee on Appropriations.
Subcommittee on Energy and Water
Development**

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uncertain or qualitative data to support decision making based on reasoned judgements With the continuing evolution of regulations for the assessment of chemical hazards and changes in thinking about how scientific decisions should be made that need is ever greater Knowledge based expert systems are well established in chemistry especially in relation to toxicology and they are used routinely to support regulatory submissions The effectiveness and continued acceptance of computer prediction depends on our ability to assess the trustworthiness of predictions and the validity of the models on which they are based Written by a pioneer in the field this book provides an essential reference for anyone interested in the uses of artificial intelligence for decision making in chemistry

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Environmental Chemistry Stanley E Manahan, 2017-02-24 With clear explanations real world examples and updated questions and answers the tenth edition of Environmental Chemistry emphasizes the concepts essential to the practice of environmental science technology and chemistry while introducing the newest innovations in the field The author follows the general format and organization popular in preceding editions including an approach based upon the five environmental spheres and the relationship of environmental chemistry to the key concepts of sustainability industrial ecology and green chemistry This readily adaptable text has been revamped to emphasize important topics such as the world water crisis It details global climate change to a greater degree than previous editions underlining the importance of abundant renewable energy in minimizing human influences on climate Environmental Chemistry is designed for a wide range of graduate and undergraduate courses in environmental chemistry environmental science and sustainability as well as serving as a general reference work for professionals in the environmental sciences and engineering

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