

CORPORATION RESOLUTION ESTABLISHING BANK ACCOUNT

with

RESOLVED, That _____, or any branch thereof, is hereby selected as a depository for the funds of this corporation, and that until it shall be otherwise notified in writing as hereinafter provided, said bank is hereby authorized to accept and receive for deposit or collection any and all items purporting to be endorsed in the corporate name of this corporation, and to make payments from the funds of this corporation on deposit with it upon and according to the check of this corporation signed by the following, and such others as may from time to time hereafter be authorized:

RESOLVED FURTHER, That the President, Vice-President, Secretary or Treasurer of this corporation, or either of them, are hereby authorized and empowered to certify and deliver to the said bank a copy of this resolution, and to execute a proper signature card, or cards, bearing the authorized signatures as aforesaid, and to execute and deliver to said bank subsequent cards bearing the specimen signatures of such others as may from time to time hereafter be authorized to sign the checks or drafts of this corporation.

RESOLVED FURTHER, That all acts and things that may be done pursuant to the authority of this resolution by the said bank, or by the officers of this corporation, are hereby fully ratified and approved, and that this resolution remain in full force and effect until written notice of revocation signed by the President, Vice-President, Secretary or Treasurer of this corporation shall have been received by said bank.

It is hereby certified that the above and foregoing is a true and correct copy of a resolution adopted by the directors of _____ of _____, at a meeting duly called and held on the _____ day of _____, _____ at which a quorum was present and voting.

IN TESTIMONY WHEREOF, Witness my signature and the seal of said corporation this the _____ day of _____.

President, Vice-President, Secretary, Treasurer*

*Only one signature required. Mark through titles of officers not signing. Be careful to affix seal of corporation.

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RM Cervero

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over

the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 °C and 64 °C respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the powdered acetamide occupies 1-2 mm of the capillary tube end c

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boiling point determination lab report conclusion objectives to correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objects can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affected by three key factors the first being intermolecular forces intermolecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 °C

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the

mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound s normal boiling point refers to its boiling point at a pressure of 760 text mm ce hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

exploring boiling points sharmin khatun chem 1141 studocu - May 17 2023

exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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dec 7 2019 1987 the sample essay on boiling point determination lab report deals with a framework of research based facts approaches and arguments concerning this theme to see the essay s introduction body paragraphs and conclusion read on should be based on observations made during the experiment

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point

at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that need to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling

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there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0 5ml 0 5 ml of material

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web the maternal health unit mah provides leadership for improving maternal and perinatal health and well being and ending preventable maternal mortality

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