

Chemistry

Metallurgy

Objective Type Questions and Answers:

- Which of the following combinations represent a metal and a non – metal respectively?
(a) lead and mercury
(b) carbon and oxygen
(c) hydrogen and nickel
(d) nickel and sulphur
- A metal which occurs native is _____
(a) nickel
(b) sodium
(c) gold
(d) mercury
- Metals which are more reactive occur in nature as _____
(a) native
(b) oxides
(c) sulphides
(d) both (b) and (c)
- Copper occurs in nature as its _____
(a) oxide
(b) chloride
(c) sulphide
(d) bromide
- Generally metals have high density. An exception is _____
(a) Cu
(b) Ag
(c) K
(d) Hg
- A metal which is not a good conductor of heat and electricity is
(a) Hg
(b) Cu
(c) Bi
(d) Ag
- An allotrope of non metal which is a good conductor of heat and electricity
(a) diamond
(b) graphite
(c) grey tin
(d) plastic sulphur
- Metals are generally lustrous. As example of a non metal which is lustrous is _____
(a) Carbon
(b) Gold
(c) Iodine
(d) Mercury

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J Elliott



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