

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
Standards and Technology
U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data					2
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd					
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																									
speed of light in vacuum																		c		299 792 458 m s ⁻¹ (exact)					
Planck constant																		h		6.626 070 15 × 10 ⁻³⁴ J s (exact)					
elementary charge																		e		1.602 176 634 × 10 ⁻¹⁹ C					
electron mass																		m _e		9.109 383 56 × 10 ⁻³¹ kg					
																		m _e /m _p		5.485 799 094 6 × 10 ⁻⁴					
proton mass																		m _p		1.672 621 9 × 10 ⁻²⁷ kg					
fine-structure constant																		α		1/137.035 999 084					
Rydberg constant																		R _∞		10 973 731.568 162 854 m ⁻¹					
																		R _H		10 967 758.340 001 421 m ⁻¹					
																		R _∞ /c		3.289 841 96 × 10 ¹⁵ Hz					
																		R _H /c		13 008 418.68 m ⁻¹					
Schrödinger constant																		k		1.380 658 × 10 ⁻²³ J K ⁻¹					
																									</

Atomic Number: 58
Ground-state Ionization Energy (eV): 5.5337

Symbol: **Ce**
Name: Cerium
Atomic Weight: 140.116
Ground-state Configuration: [Xe]4f¹5d¹6s²

*Based upon ¹²C. () indicates the mass number of the longest-lived isotope.

For a description of the data, visit physics.nist.gov/data

NIST SP 966 (September 2010)

Periodix Table Study Packet

Arturo Cuomo



Periodix Table Study Packet:

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Table of Contents Periodix Table Study Packet

1. Understanding the eBook Periodix Table Study Packet
 - The Rise of Digital Reading Periodix Table Study Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodix Table Study Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodix Table Study Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodix Table Study Packet
 - Personalized Recommendations

- Periodix Table Study Packet User Reviews and Ratings
- Periodix Table Study Packet and Bestseller Lists
- 5. Accessing Periodix Table Study Packet Free and Paid eBooks
 - Periodix Table Study Packet Public Domain eBooks
 - Periodix Table Study Packet eBook Subscription Services
 - Periodix Table Study Packet Budget-Friendly Options
- 6. Navigating Periodix Table Study Packet eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodix Table Study Packet Compatibility with Devices
 - Periodix Table Study Packet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodix Table Study Packet
 - Highlighting and Note-Taking Periodix Table Study Packet
 - Interactive Elements Periodix Table Study Packet
- 8. Staying Engaged with Periodix Table Study Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodix Table Study Packet
- 9. Balancing eBooks and Physical Books Periodix Table Study Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodix Table Study Packet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodix Table Study Packet
 - Setting Reading Goals Periodix Table Study Packet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodix Table Study Packet
 - Fact-Checking eBook Content of Periodix Table Study Packet

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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