

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $\text{C}_{10}\text{H}_{18}\text{O}_2$ (molecular)

Formula represents the number of _____ of each
in a single molecule.

Empirical $\text{C}_5\text{H}_9\text{O}$ (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit.

Atomic Mass

The average mass of atoms measured relative
to hydrogen.

Isotopes

^{12}C stable isotope abundant

^{13}C stable isotope abundant
 ^{14}C radioactive

Abundance

^{12}C abundant

^{13}C abundant

^{14}C radioactive

Note Taking Guide Episode 601 Answers Physics

Sir James Augustus Henry Murray



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