

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.0500 M NaOH 0.0500 M Acetic Acid 10.00 mL

b. 0.0500 M NaOH 0.0500 M Acetic Acid 10.00 mL

c. 0.0500 M NaOH 0.0500 M Acetic Acid 10.00 mL

Acetic Acid (pKa = 4.75) titrated with NaOH

0.0500 M NaOH 0.0500 M Acetic Acid 10.00 mL

Species	Add (M, Base (pH, or Salt (pH)	Acids with (M) (pH or pKa)	Conductivity (C or M)
HOAc	<u>0.0500</u>	<u>0.0500</u>	<u>0.0500</u>
HOAc	<u>0.0500</u>	<u>0.0500</u>	<u>0.0500</u>
HOAc	<u>0.0500</u>	<u>0.0500</u>	<u>0.0500</u>
HOAc	<u>0.0500</u>	<u>0.0500</u>	<u>0.0500</u>

Note Taking Guide Episode 1101 Answers

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