

**Table 2 – Data Collected for Nitric Acid Synthesis**

	1	2	3	4	5	6	7	8	9	10	11	12
1	10											
2	10											
3	10	10										
4	10	10	10	10	10	10	10	10	10	10	10	10
5	10	10	10	10	10	10	10	10	10	10	10	10
6	10	10	10	10	10	10	10	10	10	10	10	10
7	10	10	10	10	10	10	10	10	10	10	10	10
8	10	10	10	10	10	10	10	10	10	10	10	10
9	10	10	10	10	10	10	10	10	10	10	10	10
10	10	10	10	10	10	10	10	10	10	10	10	10
11	10	10	10	10	10	10	10	10	10	10	10	10
12	10	10	10	10	10	10	10	10	10	10	10	10

(a) Describe how you find the selected data represent the overall trend observed for the reaction that is studied in

(b) Compare the trend found in Table 2.

(c) The general rate equation for the reaction is:

Rate =

(d) Determine the rate equation for the reaction.

Rate =

(e) Which chemical species is the most reactive in the reaction?

Reaction

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