

CLO- Classification of Elements & Periodicity in Properties

MULTIPLE CHOICE QUESTIONS:

Choose the one correct option in the following:

- Identify the least stable ion amongst the following :
(a) Li^+ (b) Be^+ (c) B^+ (d) C^+
- The correct order of radii is :
(a) $\text{N} < \text{Be} < \text{B}$ (b) $\text{F}^- < \text{O}^{2-} < \text{N}^{3-}$ (c) $\text{N} < \text{Li} < \text{K}$ (d) $\text{Fe}^{2+} < \text{Fe}^{3+} < \text{Fe}^{4+}$
- Which one of the following has largest size :
(a) Na (b) Na^+ (c) Na^- (d) Li^+
- The number of elements in the 4th period of periodic table is :
(a) 8 (b) 10 (c) 18 (d) 32
- Which of the following has least electron affinity (electron gain enthalpy) :
(a) O (b) N (c) Ar (d) F
- An element of atomic number 29 belongs to which of the following block of the periodic table :
(a) s-block (b) p-block (c) d-block (d) f-block
- Which one of the following arrangements not truly represents the property indicated against it:
(a) $\text{Br}_2 < \text{Cl}_2 < \text{F}_2$ – electronegativity (b) $\text{Br}_2 < \text{F}_2 < \text{Cl}_2$ – Electron Affinity
(c) $\text{Br}_2 < \text{Cl}_2 < \text{F}_2$ – Bond energy (d) $\text{Br}_2 < \text{Cl}_2 < \text{F}_2$ – Oxidising Power
- Alkali metals in each period have :
(a) Smallest size (b) Lowest IE (c) Highest IE (d) Highest EN
- A sudden large jump between the value of second and third ionization energies of elements would be associated with which of the following electronic configuration :
(a) $2\text{p}^4 3\text{s}^1$ (b) $3\text{s}^2 3\text{p}^1$ (c) $3\text{s}^2 3\text{p}^2$ (d) $2\text{p}^4 3\text{s}^2$
- Which of the following pairs are chemically dissimilar :
(a) Na and K (b) Ba and Sr (c) Zr and Hf (d) Ca and Zn
- Collective name given to the elements with outer shell configuration $ns^2 np^6$ is :
(a) Chalkogens (b) Alkaline earths metals (c) Transition metals (d) Nobel gases
- Modern extended periodic table was given by :
(a) Bohr (b) Lothar mayer (c) Moseley (d) None alone
- The most electronegative elements is :
(a) Hydrogen (b) Fluorine (c) Chlorine (d) Oxygen
- The element having electronic configuration :
 $[\text{Kr}]4\text{d}^{10}4\text{f}^{14}5\text{s}^2 5\text{p}^6 5\text{d}^1 6\text{s}^2$ belongs to
(a) s-block (b) p-block (c) d-block (d) f-block
- An elements R forms the highest oxide R_2O_5 . R belong to :

Periodicity Multiple Choice Questions

Patrick Vollmar



Periodicity Multiple Choice Questions:

This book delves into Periodicity Multiple Choice Questions. Periodicity Multiple Choice Questions is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Periodicity Multiple Choice Questions, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Periodicity Multiple Choice Questions
 - Chapter 2: Essential Elements of Periodicity Multiple Choice Questions
 - Chapter 3: Periodicity Multiple Choice Questions in Everyday Life
 - Chapter 4: Periodicity Multiple Choice Questions in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of Periodicity Multiple Choice Questions. The first chapter will explore what Periodicity Multiple Choice Questions is, why Periodicity Multiple Choice Questions is vital, and how to effectively learn about Periodicity Multiple Choice Questions.
 3. In chapter 2, this book will delve into the foundational concepts of Periodicity Multiple Choice Questions. The second chapter will elucidate the essential principles that need to be understood to grasp Periodicity Multiple Choice Questions in its entirety.
 4. In chapter 3, the author will examine the practical applications of Periodicity Multiple Choice Questions in daily life. The third chapter will showcase real-world examples of how Periodicity Multiple Choice Questions can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Periodicity Multiple Choice Questions in specific contexts. The fourth chapter will explore how Periodicity Multiple Choice Questions is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Periodicity Multiple Choice Questions. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Periodicity Multiple Choice Questions.

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Table of Contents Periodicity Multiple Choice Questions

1. Understanding the eBook Periodicity Multiple Choice Questions
 - The Rise of Digital Reading Periodicity Multiple Choice Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodicity Multiple Choice Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodicity Multiple Choice Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodicity Multiple Choice Questions
 - Personalized Recommendations
 - Periodicity Multiple Choice Questions User Reviews and Ratings
 - Periodicity Multiple Choice Questions and Bestseller Lists
5. Accessing Periodicity Multiple Choice Questions Free and Paid eBooks
 - Periodicity Multiple Choice Questions Public Domain eBooks
 - Periodicity Multiple Choice Questions eBook Subscription Services
 - Periodicity Multiple Choice Questions Budget-Friendly Options
6. Navigating Periodicity Multiple Choice Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodicity Multiple Choice Questions Compatibility with Devices
 - Periodicity Multiple Choice Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodicity Multiple Choice Questions
 - Highlighting and Note-Taking Periodicity Multiple Choice Questions
 - Interactive Elements Periodicity Multiple Choice Questions

8. Staying Engaged with Periodicity Multiple Choice Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodicity Multiple Choice Questions
9. Balancing eBooks and Physical Books Periodicity Multiple Choice Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodicity Multiple Choice Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodicity Multiple Choice Questions
 - Setting Reading Goals Periodicity Multiple Choice Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodicity Multiple Choice Questions
 - Fact-Checking eBook Content of Periodicity Multiple Choice Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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