VICTORY 2026 (Class 10th)

CHEMISTRY
Periodic Properties and Variation of Properties

DHA–02

Q1 Which of the following elements has 2 shells, and both are completely filled?
(A) Helium
(B) Neon
(C) Calcium
(D) Boron

Q2 Why group one elements are called alkali metals?
(A) When combined with water, their oxides produce acidic solutions.
(B) When their oxides and hydroxides are treated with water, they produce alkaline solutions.
(C) When their peroxides are treated with water, they produce alkaline solutions.
(D) When treated with water, their hydroxides produce acidic solutions.

Q3 Which of the following gives the correct order of atomic radii of O, F and N?
(A) O, F, N
(B) N, F, O
(C) O, N, F
(D) F, O, N

Q4 Where would you locate the element with electronic configuration 2,8 in the Modern Periodic Table?
(A) Group 8
(B) Group 2
(C) Group 18
(D) Group 10

Q5 The order of increasing atomic size is:
(A) $I < Br < Cl < F$
(B) $I > Br > Cl > F$
(C) $Br > I > Cl > F$
(D) $Cl > Br > F > I$

Q6 Newlands arranged the elements known at that time in an increasing order of their _____ masses.
(A) electrons
(B) protons
(C) atomic
(D) ions

Q7 As per Mendeleev periodic table properties of elements are periodic function of their _____
(A) Atomic number
(B) Atomic mass
(C) Molecular mass
(D) all of the above

Q8 Demerits of Mendeleev periodic table are
(A) position of Hydrogen
(B) Position of Isotopes
(C) Position of Co and Nickel
(D) all of the above

Q9 Who made modern periodic table
(A) Mendeleev
(B) John Dalton
(C) John Newland
(D) Henry Moseley

Q10 The entire modern periodic table can be divided into how many blocks
(A) 2 blocks
(B) 3 blocks
(C) 4 blocks
(D) 5 blocks

[Periodic Variation Of Properties Answer Key](#)

Terry C. Jones

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