

Test Periodic Table and Trends SCH3U

Name: _____

Multiple Choice: Select the most correct answer (20K)

1. The mass number of an atom is determined by:

- a. the number of protons only
- b. adding the neutrons and protons
- c. adding the neutrons and electrons
- d. adding the protons and electrons



2. The letter "X" in the diagram above marks the:

- a. Region of greatest density
- b. Location of the protons
- c. Location of the neutrons
- d. Location of the electrons
- e. Nucleus

3. Most of the mass of the atom can be found in the:

- a. charges b. electron cloud c. electrons d. nucleus

4. Of the basic atomic particles, the one that would be attracted to a negatively charged metallic plate is the:

- a. proton b. neutron c. electron d. none of the above

5. Isotopes (such as hydrogen-1, hydrogen-2, and hydrogen-3) are atoms of the same element that differ in:

- a. the number of protons in the nucleus
- b. the number of electrons
- c. the atomic number
- d. the number of neutrons in the nucleus
- e. the mass number

6. Which of the following accurately summarizes the isotope Argon-40?

- a. 40 protons, 40 electrons, 18 neutrons
- b. 18 protons, 18 electrons, 22 neutrons
- c. 39.95 protons, 39.95 electrons, 21.05 neutrons
- d. 22 protons, 22 electrons, 18 neutrons

Periodic Trends Unit Test Answers

The Princeton Review



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Table of Contents Periodic Trends Unit Test Answers

1. Understanding the eBook Periodic Trends Unit Test Answers
 - The Rise of Digital Reading Periodic Trends Unit Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Unit Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Unit Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Unit Test Answers
 - Personalized Recommendations
 - Periodic Trends Unit Test Answers User Reviews and Ratings
 - Periodic Trends Unit Test Answers and Bestseller Lists

5. Accessing Periodic Trends Unit Test Answers Free and Paid eBooks
 - Periodic Trends Unit Test Answers Public Domain eBooks
 - Periodic Trends Unit Test Answers eBook Subscription Services
 - Periodic Trends Unit Test Answers Budget-Friendly Options
6. Navigating Periodic Trends Unit Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends Unit Test Answers Compatibility with Devices
 - Periodic Trends Unit Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Unit Test Answers
 - Highlighting and Note-Taking Periodic Trends Unit Test Answers
 - Interactive Elements Periodic Trends Unit Test Answers
8. Staying Engaged with Periodic Trends Unit Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Unit Test Answers
9. Balancing eBooks and Physical Books Periodic Trends Unit Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Unit Test Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Trends Unit Test Answers
 - Setting Reading Goals Periodic Trends Unit Test Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Trends Unit Test Answers
 - Fact-Checking eBook Content of Periodic Trends Unit Test Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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