PERIODIC PROPERTIES LAB

Objective: To determine the relationship between the reactivity of elements and their location on the Periodic Table.

Procedure:

Part I: Halogens

- Put 5mL of NaCl into a test tube.
- 2. Add HNO₃ to the test tube one drop. After each drop test to see if the solution turns the blue litmus paper pink. (Test this by dipping the stirring rod into the test tube and then touching it to the litmus paper).
- 3. After the blue litmus paper turns pink, add 5 more drops of HNO₃.
- 4. Add 5mL of AgNO₃ to the test tube. Record observations.
- 5. Repeat steps 1-4 for NaBr, NaI and the unknown halogen in CLEAN test tubes.

Part II: Alkaline Earth Metal Salts

- Put 1mL of each into their own test tubes: Mg(NO₃)₂, Ca(NO₃)₂, Sr(NO₃)₂, and $Ba(NO_1)_2$.
- Add 1mL of 1M H₂SO₄ to each test tube. Record observations.
- Rinse test tubes and refill with the original solutions.
- Add 5mL of 1M Na₂CO₃ to each. Record observations.
- Rinse, refill, and add 5mL of 0.25M (NH₄):C₂O₄ to each. Record observations.
- Repeat with unknown alkaline earth metal salt.

Part I: Date:

Halide	Observations	
NaCI		
NaBr		
NaI		
Unknown		

Periodic Properties Lab Answer Key

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Office of Technical Services

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unit 10 statistics this unit starts from the perspective of how we show the distribution the central tendencies and the variation within a data set it then moves onto bivariate data analysis both by hand and with the calculator lesson 1 graphical displays of data