

Chemical Formula

Talking about proportions, molecules and relative of atoms present in a compound.

Molecular $\text{C}_{10}\text{H}_{22}\text{O}_4$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical $\text{C}_5\text{H}_{11}\text{O}_2$ (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit.

Atomic Mass

The average mass of atoms present in a sample
unit.

Atomic

^{12}C mass 12 atomic mass

^{16}O mass 16 atomic mass
unit

Molecular

$\text{C}_{10}\text{H}_{22}\text{O}_4$ atomic mass

$\text{C}_{10}\text{H}_{22}\text{O}_4$ mass 222 atomic mass

$\text{C}_5\text{H}_{11}\text{O}_2$ atomic mass

Note Taking Guide Episode 601 Physics

Lei Huang

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restores it to its original state if the external force is no longer applied there are various elastic moduli such as young's
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