

Note Taking Guide: Episode 301 Name_____

Model: A familiar idea used to explain unfamiliar facts observed in nature.

Theory: An explanation of observable facts and phenomena. To

remain valid, models and theories must:

- explain all known facts
- enable scientists to make correct predictions.

Democritus:

- proposed the existence of an atom
- word comes from the greek word atomis which means not to cut or "indivisible"

Aristotle

- rejected the idea of the atom
- said matter could be cut continually

Dalton's theory proposed that atoms:

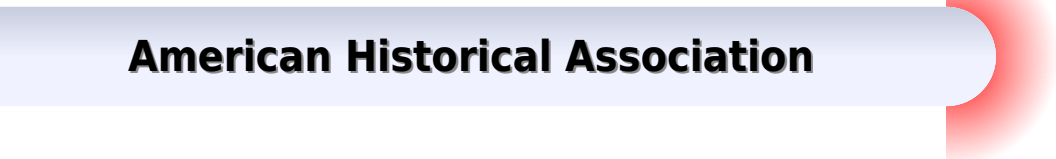
- are the building blocks of matter
- are indivisible
- of the same elements are identical
- of different elements are different
- unite in small, whole number ratios to form compounds.

J.J. Thomson

- credited with discovery of electron: a blow to Dalton's indivisible atom.
- proposed the plum pudding model of the atom:
Negatively charged electrons embedded in a ball of positive charge

Note Taking Guide Episode 301 Chemistry

American Historical Association



Note Taking Guide Episode 301 Chemistry:

Decoding **Note Taking Guide Episode 301 Chemistry**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Note Taking Guide Episode 301 Chemistry**," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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