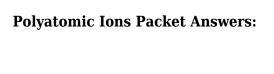
# List of Common Polyatomic Ions

Cations: Symbol	Name	Charge	Symbol	Name	Charge
NH <sub>4</sub> °	Ammonium	+1	$Hg_{2}^{-2}$	Dimercury (I)	+2
Anions:	5-4-00-0000	11.7 <u>12.4 1</u> 7.000 000 000 00			04444
Symbol	Name	Charge	Symbol	Name	Charge
$C_2H_3O_2$	Acetate	- 1	CO <sub>3</sub> ·2	Carbonate	-2
HCO <sub>3</sub>	Bicarbonate	- 1	CrO <sub>4</sub> -2	Chromate	-2
BrO <sub>3</sub>	Bromate	- 1	$Cr_2O_7^{-2}$	Dichromate	-2
BrO <sub>2</sub>	Bromite	- 1	$C_2O_4^{-2}$	Oxalate	-2
ClO <sub>4</sub>	Perchlorate	- 1	$O_2^{-2}$	Peroxide	-2
ClO <sub>3</sub>	Chlorate	- 1	SO <sub>4</sub> -2	Sulfate	-2
CIO <sub>2</sub>	Chlorite	- 1	SO <sub>3</sub> -2	Sulfite	-2
CIO	Hypochlorite	- 1	S <sub>2</sub> O <sub>3</sub> -2	Thiosulfate	-2
CN	Cyanide	- 1	SiO <sub>3</sub> 2	Silicate	-2
SCN	Thiocyanate	- 1	$MoO_4^{-2}$	Molybdate	-2
OH	Hydroxide	- 1	$WO_4^{-2}$	Tungstate	-2
NO <sub>3</sub>	Nitrate	- 1	BO <sub>3</sub> -3	Borate	-3
NO2	Nitrite	-1	$PO_4^{-3}$	Phosphate	-3
MnO <sub>4</sub>	Permanganate	- 1	$PO_3^{-3}$	Phosphite	-3
HSO.	Bisulfate	- 1	AsO <sub>4</sub> -3	Arsenate	-3
HSO <sub>3</sub>	Bisulfite	- 1	AsO <sub>3</sub> -3	Arsenite	-3
IO <sub>3</sub>	Iodate	- 1			
IO4	Periodate	- 1			
C <sub>6</sub> H <sub>5</sub> COO	Benzoate	-1			

Symbol	Name	Charge
$H_{2}PO_{4}^{-1}$	Dihydrogen phosphate	-1
$HPO_4^{-2}$	Hydrogen phosphate	-2
Fe(CN) <sub>6</sub> -3	Hexacyannoferrate (III)	-3
Fe(CN) <sub>6</sub> -4	Hexacyannoferrate (II)	-4

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**AW Chickering** 



#### Polyatomic Ions Packet Answers Book Review: Unveiling the Power of Words

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