Polyatomic Ions

Can a group of atoms have a charge?

Why?

Do you know you eat a lot of "-ates"? Next time you look at a food label, read the ingredients and you will likely find a number of ingredients that end with "-ate," such as sodium phosphate or calcium carbonate. Did you ever wonder what the chemical formulas of these ingredients look like? In this activity we will explore polyatomic ions, which are groups of atoms that carry a charge. These ions are found in our food ingredients, natural waterways, and many other chemical compounds you encounter every day.

Model 1 - Types of Ions

Monatomic Ions	Nitride N	Sulfide S	Chloride Cl)1-	
Polyatomic Ions	Nitrate	Sulfate	Ammonium H N H	
Jons	Nitrite Nitrite	Sulfite	Hydroxide	

1. Use Model 1 to complete the table below.

Name of-lon	Nitride	Nitrate	Sulfate	Sulfite	Ammonium
Charge on Ion	-3	-1	-2	-2	+1
Type and Number of Atoms	I nitrogen	1 nitrogen 3 oxygen	1 sulfur 4 oxygen	1 sulfur 3 oxygen	1 nitrogen 4 hydrogen
Chemical Formula	N3-	NO,1-	SO ₄ -	SO,2-	NH,'·

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LL Leslie

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Embracing the Beat of Expression: An Mental Symphony within Polyatomic Ions Pogil Page 3

In a global taken by displays and the ceaseless chatter of instantaneous connection, the melodic splendor and emotional symphony produced by the published term usually fade into the background, eclipsed by the constant sound and disturbances that permeate our lives. However, set within the pages of **Polyatomic Ions Pogil Page 3** a marvelous literary treasure filled with fresh feelings, lies an immersive symphony waiting to be embraced. Crafted by an elegant composer of language, that charming masterpiece conducts visitors on an emotional trip, well unraveling the hidden tunes and profound affect resonating within each carefully crafted phrase. Within the depths with this touching assessment, we shall examine the book is central harmonies, analyze their enthralling writing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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