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web the heat of reaction Δ hsoln is written after the products in units of kj mol rxn for example sodium hydroxide dissolves exothermically releasing 44 2 kilojoules per mole dissolved this process is represented as naoh s na aq oh aq Δ hsoln 44 2 kj mol rxn write equations to similarly represent the dissolving process for

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salt carbon polypropylene vermiculite water and iron

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution Δ hsoln or the amount of energy released or absorbed in kj mol of the substance being dissolved

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