## Naming Compounds Containing Polyatomic Ions

Use the following list of polyatomic ions to name the compounds given.

	OH NO <sub>2</sub> C <sub>2</sub> H <sub>2</sub> O <sub>2</sub>	hydroxide nitrate acetate	CIO <sub>3</sub> HCO <sub>3</sub> SO <sub>4</sub> CO <sub>3</sub>	chlorate bicarbonate sulfate carbonate	SO, PO, NH,	sulfite phosphate ammonium
1.	CuSO			26. NaOH		
22.	KCIO,			27. NaNO,		
3.	NH4OH					
4.	K2CO3					
5.	Na <sub>3</sub> SO,					
6.	KC2H2O2					
7.	NH,Cl					
8.	Mg(OH)2					
9.	$Zn(NO_a)_2$		v. 189 5			
10.	MgSO					
II.	CaCO <sub>2</sub>					
12.	кон					
13.	Ca(OH) <sub>2</sub>					
1.4.	K <sub>2</sub> CO <sub>3</sub>					
15.	AIPO					
16.	Na <sub>3</sub> PO <sub>4</sub>					
17.	MgSO,					
18.	AgNO <sub>3</sub>					
19.	Na <sub>2</sub> CO <sub>5</sub>			44. NiOH		
				45. ZnSO.		
21.	Ca(NO <sub>3</sub> ) <sub>2</sub>					
22.	BaSO,					
24.	LiNO <sub>3</sub>					

## **Polyatomic Ions Naming Answer**

**K Morrison** 

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