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proton

- heavy
- positively charged
- nucleon

neutron

- heavy
- neutral
- nucleon

electron

- very small (1/1800 of proton)
- negative
- outside of nucleus

properties of nucleus

- # of protons determines the element
- held together by nuclear force
- # of protons is extremely stable unlike # of electrons

atomic number

equal to the # of protons

atomic mass

of protons + # of neutrons averaged across all of the element on earth

isotopes

- when elements have different masses but the same chemical properties
- ex. silver-107 and silver-109

lewis dot

- a 2-D model that represents covalent bonds as straight lines, ionic bonds as spaces, and unbonded valence electrons as dots
- can have single, double, or triple bonds

physical properties of compounds and molecules

- melting point
- boiling point
- solubility
- electrical conductivity

molecular mass

of atoms(atomic mass)=molecular mass

Sdsu Chemistry Placement Test Study Guide

J Ma



Sdsu Chemistry Placement Test Study Guide:

Enjoying the Tune of Phrase: An Emotional Symphony within **Sdsu Chemistry Placement Test Study Guide**

In a world consumed by displays and the ceaseless chatter of fast communication, the melodic splendor and emotional symphony produced by the published term frequently fade into the back ground, eclipsed by the relentless noise and interruptions that permeate our lives. Nevertheless, situated within the pages of **Sdsu Chemistry Placement Test Study Guide** a charming fictional value overflowing with organic feelings, lies an immersive symphony waiting to be embraced. Crafted by a masterful composer of language, this charming masterpiece conducts viewers on an emotional journey, skillfully unraveling the hidden songs and profound influence resonating within each carefully constructed phrase. Within the depths of the emotional assessment, we shall discover the book is main harmonies, analyze its enthralling writing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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