



Subject Code : 0811

**WINTER – 12 EXAMINATION**  
**Model Answer**

Page No : 02/ N

- 3) **Synergistic and antagonistic effect:-** When two interacting drugs have similar action and when administered together the resultant effects is the sum of individual effect, it is termed as addition.

For eg. Combination of Sympathomimetic drugs

Amphetamine Sulphate and ephedrine Sulphate-

When two drugs having opposing pharmacological effect are prescribed together antagonism occur

For eg. Co-administration of CNS stimulants with CNS depressants.

- 4) **Contraindications:-** Some drugs are not prescribed or not taken in specific physiological or pathological conditions.

For eg. Aspirin is contraindicated if peptic ulcers are present.

D) [1 mark for definition & 1 mark for part of prescription]

**Prescription:-** Prescription is written order from a registered medical practitioner or other properly licensed practitioners, such as dentist, veterinarian etc to a pharmacist to compound and dispense a specific medication for the patients.

**Parts of Prescriptions:-**

- a) Superscription- name, address, registration number of prescriber along with name, age address on patient. Also includes symbol Rx
- b) Inscription
- c) Subscription
- d) Signature

E) White Vaseline is obtained from yellow soft paraffin by bleaching. White Vaseline is not used in ophthalmic ointment because it may contain small traces of bleaching agent which are left over after bleaching the yellow soft paraffin. Hence white Vaseline may cause irritation to eye.

(2 marks)

F) Each definition [1 mark for each definition]

**Mascara:-** Mascara is black pigmented preparation for application to the eye lashes or eyebrow to beautify the eyes. It is used to darken eyelashes & to increase their apparent length.

# Sample Question Paper For G Scheme

**S Nieto**



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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton has a h<sup>+</sup> strong acids e g sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the  $K_a$  or  $K_b$  from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web 1 approximately 3 g of hydrated oxalic acid  $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$  was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution pH is due to the acid ionization of  $\text{HCl}$  because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then  $\text{pH} = -\log 0.100 = 1.000$  pH log 0.100 = 1.000 b titrant volume 12.50 ml

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