

Assessment

Chapter Test B**Teacher Notes and Answers****8** *Chemical Equations and Reactions***TEST B**

1. b
2. a
3. b
4. d
5. a
6. b
7. d
8. a
9. coefficient
10. liquid
11. reversible reaction
12. $A + X \rightarrow AX$
13. $AX \rightarrow A + X$
14. $A + BX \rightarrow AX + B$
15. $AX + BY \rightarrow AY + BX$
16. metal hydroxides
17. combustion reaction
18. electrolysis
19. activity series
20. chemical equation
21. conservation of mass
22. gaseous
23. heated
24. A word equation and a formula equation are both qualitative. Although they both show the reactants and products of a chemical reaction, they give no information about amounts of reactants or products. A chemical equation, however, does show the relative amounts of the reactants and products in a chemical reaction.
25. Three indications of a chemical reaction are the radiation of heat and light, the production of a gas, and the formation of a precipitate.
26. zinc sulfide + oxygen $\rightarrow$
zinc oxide + sulfur dioxide
27. sodium iodide + chlorine $\rightarrow$
sodium chloride + iodine
28. $Mg + O_2 \rightarrow MgO$
29. $CaCO_3 \rightarrow CaO + CO_2$
30. decomposition reaction
31. single-displacement reaction
32. synthesis reaction
33. double-displacement reaction
34. $2AgNO_3 + CuCl_2 \rightarrow 2AgCl + Cu(NO_3)_2$
35. $2PbO_2 \rightarrow 2PbO + O_2$
36. $Zn(OH)_2 + 2CH_3COOH \rightarrow$
 $Zn(CH_3COO)_2 + 2H_2O$
37. $Mg(s) + 2H_2O(g) \rightarrow Mg(OH)_2(aq) + H_2(g)$
38. no reaction
39. $Cl_2(g) + MgBr_2(aq) \rightarrow MgCl_2(aq) + Br_2(l)$
40. $Zn(s) + 2HCl(aq) \rightarrow ZnCl_2(aq) + H_2(g)$
41. no reaction
42. $H_2(s) + CuCl_2(aq) \rightarrow H_2Cl_2(aq) + Cu(s)$
43. no reaction
44. $Mg(s) + Co(NO_3)_2(aq) \rightarrow$
 $Mg(NO_3)_2(aq) + Co(s)$

Practice Problems Reading And Study Workbook

Chapter 8

Ben Piper, David Clinton



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learn and provides the lens through which they can view learning activities and measure their intellectual growth Next she presents a specific study system that can quickly empower students to maximize their learning Then she addresses the importance of dealing with emotion attitudes and motivation by suggesting ways to change students mindsets about ability and by providing a range of strategies to boost motivation and learning finally she offers guidance to faculty on partnering with campus learning centers She pays particular attention to academically unprepared students noting that the strategies she offers for this particular population are equally beneficial for all students While stressing that there are many ways to teach effectively and that readers can be flexible in picking and choosing among the strategies she presents Sandra McGuire offers the reader a step by step process for delivering the key messages of the book to students in as little as 50 minutes Free online supplements provide three slide sets and a sample video lecture This book is written primarily for faculty but will be equally useful for TAs tutors and learning center professionals For readers with no background in education or cognitive psychology the book avoids jargon and esoteric theory

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