



NATIONAL
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2
(CHEMISTRY)
EXEMPLAR

MARKS: 100
TIME: 1 hour



The question paper consists of 10 pages, including 4 extra sheets.
Use blue or black ink for writing. Do not use correction fluid.

QUESTION 3

PHYSICAL SCIENCES PAPER 2

INTERMOLECULAR FORCES

GRADE 11

NOVEMBER 2020

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QUESTION 3 starts on a NEW page.

3.1 The boiling points in the table below were obtained during an investigation carried out to measure the boiling points of pure 1-hexanol. The table also lists the ΔH_{vap} and ΔH_{fusion} for 1-hexanol in kJ mol⁻¹.

Compound	Molecular mass (M _r)	Boiling point (°C)
A	102	16,2
B	102	21

- 3.1.1 Define the term boiling point. (2)
- 3.1.2 Is pure 1-hexanol only composed of A and B, or A, B and C, and D, one of which is present? (2)
- 3.1.3 Name the type of intermolecular force that exists between molecules of both compounds A and B and is the prime reason of their volatility. (2)
- 3.1.4 What factor of the compounds A and B, has a profound impact upon their volatility factor? (2)
- 3.1.5 Which compound (A or B) will have the higher vapour pressure? (2)
- 3.1.6 Give a reason for your answer. (2)
- 3.1.7 Which compound (A or B) will have the higher surface tension? (2)
- 3.1.8 Give a reason for the answer to 3.1.7, referring to the space for 3.1.6. (2)

Compound C (M_r 102), left alone in the table, has a smaller molecular mass than both compounds A and B but has a relatively higher boiling point of 164 °C.

- 3.1.8 Explain why the boiling point of compound C is a VAPORISING factor, as compared to compounds A and B, referring to the ΔH_{vap} and intermolecular or intramolecular forces involved. (2)

3.2 Add column C to the table in Table 1, showing all vapour pressure (kPa) and water in kPa, and also the value of the degree factor.

Compound	Boiling point (°C)	ΔH_{vap} (kJ mol ⁻¹)	ΔH_{fusion} (kJ mol ⁻¹)
A	16,2	16,2	16,2
B	21	21	21

- 3.2.1 Is there any difference in the ΔH_{vap} of the two compounds? (2)
- 3.2.2 Explain the answer to QUESTION 3.1.1 about the volatility of the compounds from Table 1. (2)

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JG Myers



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