



NATIONAL  
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2  
(CHEMISTRY)  
EXEMPLAR

MARKS: 100  
TIME: 1 hour



The question paper consists of 10 pages, including 4 extra sheets.  
Use permanent marker for writing and drawing.

# QUESTION 3

## PHYSICAL SCIENCES PAPER 2

## INTERMOLECULAR FORCES

**GRADE 11**

**NOVEMBER 2020**

Examiner's use: \_\_\_\_\_ Physical Sciences P2 \_\_\_\_\_

QUESTION 3 starts on a NEW page.

3.1 The boiling points in the table below were obtained during an investigation carried out to measure the boiling points of pure 1-hexanol. The table also lists the  $\Delta H_{\text{vap}}$  and  $\Delta H_{\text{fus}}$  for 1-hexanol in kJ mol<sup>-1</sup> and in  $^{\circ}\text{C}$  respectively.

| Compound | Molecular mass (M <sub>r</sub> ) | Boiling point (T <sub>b</sub> ) |
|----------|----------------------------------|---------------------------------|
| A        | 102                              | 162                             |
| B        | 102                              | 21                              |

- 3.1.1 Define the term boiling point. (2)
- 3.1.2 Is pure 1-hexanol only composed of A and B? at 0  $^{\circ}\text{C}$  and 101 kPa of external pressure? (2)
- 3.1.3 Name the type of intermolecular force that exists between molecules of both compounds A and B and its relative value of these molecules. (2)
- 3.1.4 What kind of the compound A or B is less (1-hexanol) under these proposed trends? (2)
- 3.1.5 What compound B or A is less the higher vapor pressure? (2)
- 3.1.6 What compound B or A is less the higher vapor pressure? (2)

Compound B (M<sub>r</sub> 102) is shown in the table. Use a suitable reference data that both compounds A and B for the  $\Delta H_{\text{vap}}$  and boiling point of 102  $^{\circ}\text{C}$ .

- 3.1.6 Explain why the boiling point of compound B is (1-hexanol) higher than compound A and B for the  $\Delta H_{\text{vap}}$  and intermolecular forces. (2)

3.2 Add table 3.2 is added to each column of pure substances (100) are shown in separate columns on sheet of the diagram below.



- 3.2.1 Is solid 100 (100) or 100, at the table shown? (2)
- 3.2.2 Explain the process to (100) 100 about the melting in the intermolecular forces. (2)

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# Physical Science Paper2 November 2014

**Lei Huang**



**Physical Science Paper2 November 2014:**

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