

Periodic Trends Worksheet

Directions: Use the trends in electronegativity to answer the questions.

1. Write the following elements by increasing atomic radius: sodium, sulfur, chlorine, oxygen, potassium.

K, Na, S, Cl, O

2. Write the following elements by increasing electronegativity: sodium, oxygen, sulfur, chlorine, potassium.

O, Cl, S, Na, K

3. Which atom has a higher electronegativity: fluorine or sulfur?

Fluorine is more electronegative than sulfur because it is further to the right.

4. Which two elements in the same family generally have similar properties?

There have been two elements that share a strong similarity with potassium.

5. Arrange the following properties in order from left to right across the periodic table:

- a. atomic radius (increases to the right) **increases**
- b. first ionization energy **increases**
- c. electronegativity **increases**

6. Which period in the periodic table shows atoms with the greatest atomic radius? Which element has the greatest?

Period 7, francium, cesium, barium, and thallium.

7. Which period in the periodic table shows atoms with the smallest atomic radius? Which element has the smallest?

Period 1, hydrogen, helium, lithium, beryllium, boron, carbon, nitrogen, oxygen, fluorine, neon.

8. Write the atoms in order from the lowest to the highest electronegativity:

a. **Li**
 b. **Na**
 c. **K**
 d. **Rb**
 e. **Fr**
 f. **Cs**
 g. **Ba**
 h. **La**
 i. **Be**
 j. **Mg**
 k. **Ca**
 l. **Sc**
 m. **Y**
 n. **Zr**
 o. **Nb**
 p. **Ta**
 q. **Hf**
 r. **Ta**
 s. **Pb**
 t. **Bi**
 u. **Po**
 v. **At**
 w. **Fr**
 x. **Ra**
 y. **Ac**
 z. **Th**
 aa. **Pa**
 ab. **U**
 ac. **Np**
 ad. **Pu**
 ae. **Am**
 af. **Cm**
 ag. **Bk**
 ah. **Cf**
 ai. **Es**
 aj. **Fm**
 ak. **Md**
 al. **No**
 am. **Lr**
 an. **Lu**
 ao. **Hf**
 ap. **Ta**
 aq. **W**
 ar. **Re**
 as. **Os**
 at. **Ir**
 au. **Pt**
 av. **Au**
 aw. **Hg**
 ax. **Tl**
 ay. **Pb**
 az. **Bi**
 ba. **Po**
 bb. **At**
 bc. **Fr**
 bd. **Ra**
 be. **Ac**
 bf. **Th**
 bg. **Pa**
 bh. **U**
 bi. **Np**
 bj. **Pu**
 bk. **Am**
 bl. **Cm**
 bm. **Bk**
 bn. **Cf**
 bo. **Es**
 bp. **Fm**
 bq. **Md**
 br. **No**
 bs. **Lr**
 bt. **Lu**
 bu. **Hf**
 bv. **Ta**
 bw. **W**
 bx. **Re**
 by. **Os**
 bz. **Ir**
 ca. **Pt**
 cb. **Au**
 cc. **Hg**
 cd. **Tl**
 ce. **Pb**
 cf. **Bi**
 cg. **Po**
 ch. **At**
 ci. **Fr**
 cj. **Ra**
 ck. **Ac**
 cl. **Th**
 cm. **Pa**
 cn. **U**
 co. **Np**
 cp. **Pu**
 cq. **Am**
 cr. **Cm**
 cs. **Bk**
 ct. **Cf**
 cu. **Es**
 cv. **Fm**
 cw. **Md**
 cx. **No**
 cy. **Lr**
 cz. **Lu**
 da. **Hf**
 db. **Ta**
 dc. **W**
 dd. **Re**
 de. **Os**
 df. **Ir**
 dg. **Pt**
 dh. **Au**
 di. **Hg**
 dj. **Tl**
 dk. **Pb**
 dl. **Bi**
 dm. **Po**
 dn. **At**
 do. **Fr**
 dp. **Ra**
 dq. **Ac**
 dr. **Th**
 ds. **Pa**
 dt. **U**
 du. **Np**
 dv. **Pu**
 dv. **Am**
 dw. **Cm**
 dx. **Bk**
 dy. **Cf**
 dz. **Es**
 ea. **Fm**
 eb. **Md**
 ec. **No**
 ed. **Lr**
 ee. **Lu**
 ef. **Hf**
 ef. **Ta**
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Sabine Zange



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