

TYBSC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

1	Which of the following is a uni-univalent electrolyte? a) KCl b) CuCl_2 c) AlCl_3 d) MgCl_2
2	The ionic strength of a solution of molality m_i and valency of the ion Z_i is ____ a) $\mu = \frac{1}{2} m_i Z_i^2$ b) $\mu = \sum m_i Z_i^2$ c) $\mu = \frac{1}{2} \sum Z_i^2$ d) $\mu = \frac{1}{2} \sum m_i Z_i^2$
3	Which of the following is an example of electrolyte concentration cell with transference? a) $\text{Cu} \text{CuSO}_4 \text{AgNO}_3 \text{Ag}$ b) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)} \text{AgCl}_{(s)} \text{Ag}$ c) $\text{Pt}, \text{H}_{2(g)} \text{HCl}_{(aq)}(a_1) \text{AgNO}_3(a_2) \text{Ag}$ d) $\text{Ag} - \text{AgCl} \text{NaCl}(a_1) \text{NaCl}(a_2) \text{AgCl} \cdot \text{Ag}$
4	A concentration cell which has salt bridge in its representation is ____ a) electrolyte concentration cell with transference b) Electrode concentration cell c) Electrolyte concentration cell without transference d) Gas concentration cell
5	The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by..... a) $E_j = E_{\text{oxidation}} + E_{\text{reduction}}$ b) $E_{\text{cell}} = E_{\text{oxidation}} + E_{\text{reduction}} + E_j$ c) $E_{\text{cell}} = E_{\text{oxidation}} - E_{\text{reduction}} - E_j$ d) $E_j = E_{\text{oxidation}} - E_{\text{reduction}}$
6	The activity 'a' is related to the activity coefficient ' γ ' by the equation____ a) $a = \frac{m}{\gamma}$ b) $\gamma = a - m$ c) $a = m + \gamma$ d) $a = m\gamma$

Physical Chemistry Multiple Choice Questions

M.M.L. Oblitas



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Physical Chemistry: A Level Multiple Choice Questions Margaret Marion Langley Oblitas, 1970-01-01

Physical Chemistry M.M.L. Oblitas, 1971

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The Pearson Guide to Physical Chemistry for the IIT JEE Singhal, Sharma K.K. & Sharma L.K., 2016

A Textbook of Physical Chemistry Sharma K.K. & Sharma L.K., 2016

A Textbook of Physical Chemistry, 6th Edition Atkins, Julio De Paula, James Keeler, 2023

Revised edition of Atkins Physical chemistry Peter Atkins Julio de Paula James Keeler Eleventh edition 2018

Elements of Physical Chemistry Peter Atkins, Julio de Paula, 2013

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AQA A-level Year 2 Chemistry Student Guide: Physical chemistry 2 Alyn G. McFarland, Nora Henry, 2016-04-04 Exam Board AQA Level A level Subject Chemistry First Teaching September 2016 First Exam June 2017 Written by experienced examiners Alyn McFarland and Nora Henry this Student Guide for Chemistry Helps students identify what they need to know with a concise summary of the topics examined in the AS and A level specifications Consolidates understanding with tips and knowledge check questions Provides opportunities to improve exam technique with sample answers to exam style questions

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