

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

NATIONAL
SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

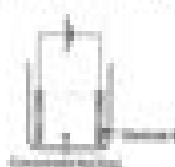
The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reagent used in reaction IV. (1 mark)



4.8 Draw the half-cell reaction for the zinc electrode. (1 mark)

4.9 Write the overall cell reaction. (1 mark)

4.10 Explain the reason for the change in the concentration of the reactants in the reaction.



4.11 Explain the reason for the change in the concentration of the reactants in the reaction.

4.12 The simplified diagram below represents an electrochemical cell. The electrodes are made of copper.

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Puja Mehta



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Table of Contents Physical Science Paper 2 November 2013 Memorandum

1. Understanding the eBook Physical Science Paper 2 November 2013 Memorandum
 - The Rise of Digital Reading Physical Science Paper 2 November 2013 Memorandum
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper 2 November 2013 Memorandum
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Paper 2 November 2013 Memorandum
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Paper 2 November 2013 Memorandum
 - Personalized Recommendations
 - Physical Science Paper 2 November 2013 Memorandum User Reviews and Ratings
 - Physical Science Paper 2 November 2013 Memorandum and Bestseller Lists
5. Accessing Physical Science Paper 2 November 2013 Memorandum Free and Paid eBooks
 - Physical Science Paper 2 November 2013 Memorandum Public Domain eBooks
 - Physical Science Paper 2 November 2013 Memorandum eBook Subscription Services
 - Physical Science Paper 2 November 2013 Memorandum Budget-Friendly Options
6. Navigating Physical Science Paper 2 November 2013 Memorandum eBook Formats

- ePub, PDF, MOBI, and More
- Physical Science Paper 2 November 2013 Memorandum Compatibility with Devices
- Physical Science Paper 2 November 2013 Memorandum Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Paper 2 November 2013 Memorandum
 - Highlighting and Note-Taking Physical Science Paper 2 November 2013 Memorandum
 - Interactive Elements Physical Science Paper 2 November 2013 Memorandum
- 8. Staying Engaged with Physical Science Paper 2 November 2013 Memorandum
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Paper 2 November 2013 Memorandum
- 9. Balancing eBooks and Physical Books Physical Science Paper 2 November 2013 Memorandum
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Paper 2 November 2013 Memorandum
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Paper 2 November 2013 Memorandum
 - Setting Reading Goals Physical Science Paper 2 November 2013 Memorandum
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Paper 2 November 2013 Memorandum
 - Fact-Checking eBook Content of Physical Science Paper 2 November 2013 Memorandum
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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