

- Answer ALL the questions in the same handwriting
- Non-programmable calculator may be used
- Appropriate mathematical instruments may be used
- Round off your final numerical answers to a minimum of (5) decimal places
- Give detailed reasons, discussions, if asked where needed
- Write neatly and legibly.

Learner's Name

Learner's Surname

Grade

Page 1 of 5

The following are represented by A, B and C:
 A) Amplitude, wavelength, wave length
 B) Crest, trough, wavelength
 C) Amplitude, wave length, pulse length
 D) Crest, trough, wavelength

[2 Marks]

1.5. The displacement of mass m at points A and B is unknown.



Page 2 of 5

Physical Sciences – Paper 1 (Physical)

How many wavelengths separate point A and B?

A) 1,25

B) 1

C) 1,5

D) 3

[2 Marks]

1.6. Devices are sometimes placed under the headlights of a car to detect if a car approaches. This equipment uses the following wave...

A) X-ray

B) infrared light

C) Ultrasound

D) Ultra violet light

[2 Marks]

1.7. The following are an example of electromagnetic waves:

A) Microwaves, visible light, X-rays

B) X-rays, visible light, microwaves

C) Microwaves, X-rays, visible light

Physical Sciences – Paper 1 (Physical)

2.2. Identify TWO points in the graph at which the particles would be moving in the same direction

[2 Marks]

2.3. Which point has the maximum displacement?

[1 Mark]

Calculate

2.4. The frequency of the wave produced

[2 Marks]

2.5. The wavelength of the wave produced

[1 Mark]

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web oct 31 2023 an isotope is one of two or more species of atoms of a chemical element with the same atomic number and position in the periodic table and nearly identical chemical behavior but with different atomic masses and physical properties every chemical element has one or more isotopes

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web jul 18 2018 answer link due to the number of neutrons isotopes differ with respect to the number of neutrons their nuclei contain where neutrons are massive particles of zero charge a nucleus may contain variable numbers of neutrons and this gives rise to the phenomenon of isotopes

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web isotopes are distinct nuclear species or nuclides as technical term of the same chemical element they have the same atomic number number of protons in their nuclei and position in the periodic table and hence belong to the same chemical element but differ in nucleon numbers mass numbers due to different numbers of neutrons in

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope's mass number

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web nov 27 2016 explanation basically isotopes are nuclei that have the same number of protons and are thus the same element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn't differ all that much but its mass does differ thus its density is a little

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web chemistry questions and answers isotopes differ from each other in what ways a they have different numbers of electrons outside the nucleus b they have different atomic number o c they have different numbers of neutrons in the nucleus d more than one response is correct e they have different numbers of protons in the nucleus