

| Element    | symbol | Atomic number | Electronic arrangement in different orbit |   |   |   |   |   |
|------------|--------|---------------|-------------------------------------------|---|---|---|---|---|
|            |        |               | K                                         | L | M | N | O | P |
| Hydrogen   | H      | 1             | 1                                         |   |   |   |   |   |
| Helium     | He     | 2             | 2                                         |   |   |   |   |   |
| Lithium    | Li     | 3             | 2                                         | 1 |   |   |   |   |
| Beryllium  | Be     | 4             | 2                                         | 2 |   |   |   |   |
| Boron      | B      | 5             | 2                                         | 3 |   |   |   |   |
| Carbon     | C      | 6             | 2                                         | 4 |   |   |   |   |
| Nitrogen   | N      | 7             | 2                                         | 5 |   |   |   |   |
| Oxygen     | O      | 8             | 2                                         | 6 |   |   |   |   |
| Fluorine   | F      | 9             | 2                                         | 7 |   |   |   |   |
| Neon       | Ne     | 10            | 2                                         | 8 |   |   |   |   |
| Sodium     | Na     | 11            | 2                                         | 8 | 1 |   |   |   |
| Magnesium  | Mg     | 12            | 2                                         | 8 | 2 |   |   |   |
| Aluminium  | Al     | 13            | 2                                         | 8 | 3 |   |   |   |
| Silicon    | Si     | 14            | 2                                         | 8 | 4 |   |   |   |
| Phosphorus | P      | 15            | 2                                         | 8 | 5 |   |   |   |
| Sulphur    | S      | 16            | 2                                         | 8 | 6 |   |   |   |
| Chlorine   | Cl     | 17            | 2                                         | 8 | 7 |   |   |   |
| Argon      | Ar     | 18            | 2                                         | 8 | 8 |   |   |   |
| Potassium  | K      | 19            | 2                                         | 8 | 8 | 1 |   |   |
| Calcium    | Ca     | 20            | 2                                         | 8 | 8 | 2 |   |   |

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●  $c = v\lambda$

●  $\bar{v} = \frac{1}{\lambda} = \frac{v}{c}$

[ $c$  = velocity of light in free space,

$v$  = frequency,  $\lambda$  = wavelength]

[ $\bar{v}$  = wave number]

# Physical Science Notes Chapter 4 Atomic Structure

**M Planty**



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