Use your notebook to complete all of the following review questions. Please note that your completed notebook, corrected quizzes and unit review sheets are the <u>BEST</u> study tools. These questions will only help to reinforce some of the exam topics.

1. Draw a line continuum to show the metric prefixes in order from largest (on the left) to smallest (on the right).



- Complete the following metric conversions:

  - a) 75 g = \_\_0.075\_\_ kg e) 4.2 kL = \_420 000\_ cL
  - b) 4530 cm = 0.0453 km
- f) 83.2 cm = 832 mm g) 975 mg = \_0.975\_ g
- c) 0.8 A = 800 mA d) 3200 mA = \_\_3,2\_ A
  - h) 975 mg = 0.000975 kg
- Complete the following table using your knowledge of the Particle Theory of Matter:

STATE of MATTER	FORCES OF ATTRACTION	SPACING OF MOLECULES	MOVEMENT OF MOLECULES
solid	strongest	close together	least
liquid	weaker	farther apart	more
gas	weakest	farthest apart	most

- Changes of state are examples of \_physical\_ changes.
  - a) Complete the following diagram with labels and arrows to show all of the SIX changes of state. 3. EVAPOURATION 4, CONDENSATION



5. Name FIVE ways in which you know a chemical change has probably taken place.

colour change bubbles form precipitate forms light given off heat produced

# **Physical Science Grade 9 Final Exam**

Jianjun Gao

#### **Physical Science Grade 9 Final Exam:**

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# **Table of Contents Physical Science Grade 9 Final Exam**

- 1. Understanding the eBook Physical Science Grade 9 Final Exam
  - The Rise of Digital Reading Physical Science Grade 9 Final Exam
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Physical Science Grade 9 Final Exam
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science Grade 9 Final Exam
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Physical Science Grade 9 Final Exam
  - Personalized Recommendations
  - Physical Science Grade 9 Final Exam User Reviews and Ratings
  - Physical Science Grade 9 Final Exam and Bestseller Lists
- 5. Accessing Physical Science Grade 9 Final Exam Free and Paid eBooks
  - Physical Science Grade 9 Final Exam Public Domain eBooks
  - Physical Science Grade 9 Final Exam eBook Subscription Services
  - Physical Science Grade 9 Final Exam Budget-Friendly Options
- 6. Navigating Physical Science Grade 9 Final Exam eBook Formats
  - o ePub, PDF, MOBI, and More
  - Physical Science Grade 9 Final Exam Compatibility with Devices
  - Physical Science Grade 9 Final Exam Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science Grade 9 Final Exam
  - Highlighting and Note-Taking Physical Science Grade 9 Final Exam
  - Interactive Elements Physical Science Grade 9 Final Exam
- 8. Staying Engaged with Physical Science Grade 9 Final Exam

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Physical Science Grade 9 Final Exam
- 9. Balancing eBooks and Physical Books Physical Science Grade 9 Final Exam
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science Grade 9 Final Exam
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Grade 9 Final Exam
  - Setting Reading Goals Physical Science Grade 9 Final Exam
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Grade 9 Final Exam
  - Fact-Checking eBook Content of Physical Science Grade 9 Final Exam
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - o Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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