



Name: _____ Date: _____ Class Period: _____

Strength of Acids & Bases

Directions: The pH values of several solutions are shown below. Note: strong acids have a pH of 3 or less, and strong bases have a pH of 11 or greater.

- 1) Color the pH scale the appropriate colors to reflect the colors of litmus paper that reflect the strength of pH of solutions.
- 2) Write the name of the item on the correct location on the pH scale.
- 3) Fill in the items in the table to determine if they are a strong or weak acid or base.

pH Scale

0	1	2	3	4	5	6	7	8	9	10	11	12	13	14
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Strong Acids	Weak Acids	Strong Bases	Weak Bases

pH Values:

Milk – 6.0

Soda – 3.0

Laundry detergent – 10.5

Industrial Strength Soap – 12.5

Distilled Water – 7.0

Toothpaste – 9.0

Oven cleaner – 12.5

Sodium hydroxide – 14.0

Lemon juice – 2.4

Black coffee – 5.0

Baking soda – 8.4

Hydrochloric acid – 1.0

Vinegar – 2.5

Tomato juice – 4.0

Beach – 13.0

Physical Science 8 4 Strength Of Acids And Bases

Ian Pickup



Physical Science 8 4 Strength Of Acids And Bases:

Reviewing **Physical Science 8 4 Strength Of Acids And Bases**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Physical Science 8 4 Strength Of Acids And Bases**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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