# **EMPIRICAL & MOLECULAR FORMULA**

- In Dalton's atomic theory, he stated that atoms will combine with one another in simple, whole number ratios to form compounds.
- An example of a molecular formula (the actual formula) is C<sub>6</sub>H<sub>6</sub>. This is Benzene.
- An example of a simple whole number ratio, or the empirical formula for Benzene would be CH.

# **Pratice Writing Empirical And Molecular Formula**

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