Specific Heat

DRECTIONS: Use formula q + (m)(4/0)(C) to salve the following problems. Show all reork and units. Use any lowering tookbox down for specific heat of various substances.

- A 20g page of metal shoods: 1,200 postered heat-energy and its temperature change: from 25°C to 175°C. Calculate the specific organizated the metal. What is the Madazinetal?
- How many profes of heat are peopled to make the temperature of 15 G g of iron from 11°C to 66°C, if the operation heat of iron to 0.45 Pg*C?
- To what temperature will a 50.0 g percent bears make if it absorbed histogrades of best and its specific best capacity is 0.30 Mg*C≥. The initial temperature of the bears is 20.0°C.
- Identify the intotance of 2500.0 g of the intotance absorbs 74,300 joules of best, and its temperature changes from 25°C to 47°C.
- 100.0 g of 1.0°C vester is heated until its temperature to 30°C. If the specific heat of vester is 4.10 Fg°C, calculate the amount of heat-energy needed to cause this rise in temperature.
- 25.0 g of antisocopy is heated from 20°C to 180°C, and about to 750 joules of heat in the process. Calculate the specific heat-capacity of antisocopy.
- What is the specific heat capacity of quarts glass if 200:0 g of the glass decodes 2520 jordes of heat and the hosperstope rises 18:0°CF.
- If a rample of transium to initially at 25°C, what in the final temperature if 250.0 g of transium absorbs 1.0 inlegible of heat, and the operation heat of transium is 0.525 Fg*C?
- How rounds energy must be absorbed by 200.0 g of sheet to increase its temperature from 100:0°C to 450:0°C? (C.of sheet = 49.0 g °C)
- When 55.0 g of plattures drops to temperature from 200.0°C to 250.0°C, horrorach heat-energy or released? (C of plattures = 120 Mg °C)
- How suitch esergy is required to heat 125-0 g of lithium from 2-0°C to 30.0°C? (C of lithium = 3.6 Jul*C)
- E2: How much heart (in hd) in given out when 200 0 g of lead cools from 200,0°C to 20 0°C?

Practice Problems For Specific Heat

Kenneth W. Raymond

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