CHEM 1411 – General Chemistry I
Practice Problems, Chapters 1–3

Chapter 1 – Chemistry: The Study of Change

1. Element, compound, homogeneous mixture (solution), or heterogeneous mixture:

a) orange juice (freshly-squeezed) b) brass c) 0.9% saline (NaCl) solution
d) garden soil e) room air f) methane gas
g) sodium metal h) N_2 gas i) $Cu(NO_3)_2$ crystals

2. Define (some of these terms are found in Chapters 2 and 3):

a) element
b) compound
c) solution
d) physical change
e) chemical change
f) law of conservation of mass
g) intensive properties
h) extensive properties
i) precision
j) accuracy
k) isotopes
l) stoichiometry
m) the mole
n) Avogadro's number
o) limiting reactant

3. Convert

a) 5 feet 10 inches to meters
b) 55 miles per hour to cm per second
c) 5.0 quarts to liters
d) 1.00 ft^2 to cm^2
e) 32.06 amu to g
f) 25 g of carbon to moles of carbon
g) 1.00 g of gold to number of atoms of gold
h) 25 °F to Kelvins

4. How many significant figures are in the following numbers?

a) 3.1416 b) 6.022×10^{23} atoms/mole c) 0.000104 g
d) 6.0 g/mL e) 1.00794 amu f) 5000 mi

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