



GAUTENG PROVINCE

EDUCATION
REPUBLIC OF SOUTH AFRICA

**GAUTENG DEPARTMENT OF EDUCATION
PROVINCIAL EXAMINATION
NOVEMBER 2021**

GRADE 11

**PHYSICAL SCIENCES
(PHYSICS)**

PAPER 1

TIME: 2 hours

MARKS: 100

11 pages, an answer sheet and 2 data sheets

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web equilibrium moles $hscn$ $2.00 \times 10^{-5} \text{ mol}$ $3.00 \times 10^{-6} \text{ mol}$ $1.70 \times 10^{-5} \text{ mol}$ $hscn$ knowing the number of moles of Fe^{3+} and $hscn$ present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species $mol\ Fe^{3+}$ $1.70 \times 10^{-5} \text{ mol}$ Fe^{3+} $8.50 \times 10^{-4} \text{ M}$

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