

# Physical Science EOCT Study Guide – Section 3

## Physical Science: Force and Motion

### Standards: PS.3.1 and 3.2

1. Students will analyze cause-and-effect relationships and flow of energy within a system.
  - a. Explain thermal energy (heat) and describe the principle of conservation of energy as it applies to heat transfer in an object.
  - b. Explain heat. Thermal energy flows from objects that are hotter to objects that are cooler. Heat is the transfer of energy from one object to another.
  - c. Explain specific heat. The amount of heat energy needed to raise the temperature of 1 kg of a substance by 1°C.
  - d. What is the formula for determining that change in heat that energy can add?
 
$$Q = mc\Delta T$$

$$\Delta T = T_2 - T_1$$
  - e. A sample of 0.5 kg of water at 20°C and specific heat of 4186 J/kg°C. Find the change in heat energy if the block is 2 mmol from 10°C to 20°C.
 
$$Q = mc\Delta T$$

$$Q = (0.5 \text{ kg})(4186 \text{ J/kg°C})(20^\circ\text{C} - 10^\circ\text{C})$$

$$Q = 20,930 \text{ J}$$
  - f. Explain conduction, convection, and radiation. Give examples of each.
 Conduction is the transfer of heat through direct contact. Convection is the transfer of heat through the movement of a fluid. Radiation is the transfer of heat through electromagnetic waves.

2. Explain the following phase change diagrams.



3. Students will determine relationships among force, mass and motion.
  - a. Explain the difference between distance and displacement.
 Distance is the total length of the path traveled. Displacement is the change in position from the starting point to the ending point.
  - b. What is the difference between average speed and instantaneous speed?
 Average speed is the total distance divided by the total time. Instantaneous speed is the speed at a particular moment in time.
  - c. What is the formula for average speed?
 
$$v = \frac{d}{t}$$

$$v = \frac{d}{t}$$

# Physical Science Eoct Study Guide Energy Transformations

**R Sandford**



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