

TYBSC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

|   |                                                                                                                                                                                                                                                                                                                                                                                                                                                       |
|---|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| 1 | Which of the following is a uni-univalent electrolyte?<br>a) KCl<br>b) $\text{CuCl}_2$<br>c) $\text{AlCl}_3$<br>d) $\text{MgCl}_2$                                                                                                                                                                                                                                                                                                                    |
| 2 | The ionic strength of a solution of molality $m_i$ and valency of the ion $Z_i$ is ____<br>a) $\mu = \frac{1}{2} m_i Z_i^2$<br>b) $\mu = \sum_i m_i Z_i^2$<br>c) $\mu = \frac{1}{2} \sum_i Z_i^2$<br>d) $\mu = \frac{1}{2} \sum_i m_i Z_i^2$                                                                                                                                                                                                          |
| 3 | Which of the following is an example of electrolyte concentration cell with transference?<br><br>a) $\text{Cu}   \text{CuSO}_4    \text{AgNO}_3   \text{Ag}$<br>b) $\text{Pt}, \text{H}_{2(g)}   \text{HCl}_{(aq)}   \text{AgCl}_{(s)}   \text{Ag}$<br>c) $\text{Pt}, \text{H}_{2(g)}   \text{HCl}_{(aq)}(a_1)    \text{AgNO}_3(a_2)   \text{Ag}$<br>d) $\text{Ag} - \text{AgCl}   \text{NaCl}(a_1)   \text{NaCl}(a_2)   \text{AgCl} \cdot \text{Ag}$ |
| 4 | A concentration cell which has salt bridge in its representation is ____<br>a) electrolyte concentration cell with transference<br>b) Electrode concentration cell<br>c) Electrolyte concentration cell without transference<br>d) Gas concentration cell                                                                                                                                                                                             |
| 5 | The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by.....<br>a) $E_j = E_{\text{oxidation}} + E_{\text{reduction}}$<br>b) $E_{\text{cell}} = E_{\text{oxidation}} + E_{\text{reduction}} + E_j$<br>c) $E_{\text{cell}} = E_{\text{oxidation}} - E_{\text{reduction}} - E_j$<br>d) $E_j = E_{\text{oxidation}} - E_{\text{reduction}}$                                           |
| 6 | The activity 'a' is related to the activity coefficient ' $\gamma$ ' by the equation____<br>a) $a = \frac{m}{\gamma}$<br>b) $\gamma = a - m$<br>c) $a = m + \gamma$<br>d) $a = m\gamma$                                                                                                                                                                                                                                                               |

# Physical Chemistry Multiple Choice Questions

**Dr Kusum Baser**



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