

Density Worksheet

Physical Science

$D = m/V$

Densities of Common Substances @ 20°C				
Substance	Density (g/cm ³)		Substance	Density (g/cm ³)
Oxygen	0.00133		Aluminum	2.70
Hydrogen	0.000084		Iron	7.87
Ethanol	0.785		Copper	8.96
Benzene	0.880		Silver	10.5
Water	1.000		Lead	11.34
Magnesium	1.74		Mercury	13.6
Salt (sodium chloride)	2.16		Gold	19.32

1. The ratio of an object's mass to its _____ is called the *density* of the object.
2. A kilogram of lead occupies a much smaller volume than a kilogram of water, because _____ has a much higher *density*.
3. For the masses and volumes indicated, calculate the **density** in grams per cubic centimeters.
 - a. mass = 453 g; volume = 225 cm³
 - b. mass = 5.0 g; volume = 10.0 cm³
 - c. mass = 26.1 g; volume = 2.0 mL
4. If 89.2 mL of a liquid has a mass of 75.2 g, calculate the liquid's density.
5. A cube of metal weighs 1450 g and displaces 542 mL of water when immersed. Calculate the density of the metal.

Physical Science If8767 Page 99 Answers

Denise Eby, Robert B. Horton



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Table of Contents Physical Science If8767 Page 99 Answers

1. Understanding the eBook Physical Science If8767 Page 99 Answers
 - The Rise of Digital Reading Physical Science If8767 Page 99 Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Page 99 Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science If8767 Page 99 Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Page 99 Answers
 - Personalized Recommendations

- Physical Science If8767 Page 99 Answers User Reviews and Ratings
- Physical Science If8767 Page 99 Answers and Bestseller Lists
- 5. Accessing Physical Science If8767 Page 99 Answers Free and Paid eBooks
 - Physical Science If8767 Page 99 Answers Public Domain eBooks
 - Physical Science If8767 Page 99 Answers eBook Subscription Services
 - Physical Science If8767 Page 99 Answers Budget-Friendly Options
- 6. Navigating Physical Science If8767 Page 99 Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science If8767 Page 99 Answers Compatibility with Devices
 - Physical Science If8767 Page 99 Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science If8767 Page 99 Answers
 - Highlighting and Note-Taking Physical Science If8767 Page 99 Answers
 - Interactive Elements Physical Science If8767 Page 99 Answers
- 8. Staying Engaged with Physical Science If8767 Page 99 Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science If8767 Page 99 Answers
- 9. Balancing eBooks and Physical Books Physical Science If8767 Page 99 Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science If8767 Page 99 Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science If8767 Page 99 Answers
 - Setting Reading Goals Physical Science If8767 Page 99 Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science If8767 Page 99 Answers
 - Fact-Checking eBook Content of Physical Science If8767 Page 99 Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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