

|     |       |                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                          |     |
|-----|-------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|-----|
|     |       | <p>gerangskik <input checked="" type="checkbox"/> is om die hoeke tussen hulle te maksimeer. <input checked="" type="checkbox"/> Strukturele elektronpare is bindingspare plus alleenpare.</p> <p>OF</p> <p>Valensdop-elektronpaarafstotingsmodel is 'n model om die vorms van molekules te voorspel <input checked="" type="checkbox"/> waarin die elektronpaar vanaf die buitenste dop <input checked="" type="checkbox"/> van 'n verwysingsatoom rondom hierdie atoom gerangskik is om die afstoting tussen hulle te minimaliseer. <input checked="" type="checkbox"/></p> <p>3 marks/punte</p>                                                                                                                                                                                       |     |
|     | 1.2.4 | <p>The molecule is a tetrahedron <input checked="" type="checkbox"/></p> <p>Die molekule is 'n tetraëder <input checked="" type="checkbox"/></p> <p>1 mark/punt</p>                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                      | LO2 |
|     | 1.2.5 |  <p>Name trigonal pyramid <input checked="" type="checkbox"/></p> <p>Noem trigonale piramidele <input checked="" type="checkbox"/></p> <p>3 marks/punte</p>                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                                            | LO4 |
| 1.3 | 1.3.1 | <p>The electron pair is equally shared between the two Cl atoms <input checked="" type="checkbox"/> and the resulting Cl<sub>2</sub> molecule has no degree of polarity. <input checked="" type="checkbox"/></p> <p>Die elektronpaar word ewe veel gedeel tussen die twee Cl-atome <input checked="" type="checkbox"/> en die resulterende Cl<sub>2</sub>-molekule het geen mate van polariteit nie <input checked="" type="checkbox"/>.</p> <p>2 marks/punte</p>                                                                                                                                                                                                                                                                                                                        | LO2 |
|     | 1.3.2 | <p>The tendency <input checked="" type="checkbox"/> of an atom to pull the bonded pair <input checked="" type="checkbox"/> of electrons towards itself. <input checked="" type="checkbox"/></p> <p>Die neiging <input checked="" type="checkbox"/> van 'n atoom om die gebinde elektronpare <input checked="" type="checkbox"/> na homself te trek <input checked="" type="checkbox"/>.</p> <p>3 marks/punte</p>                                                                                                                                                                                                                                                                                                                                                                         | LO2 |
|     | 1.3.3 | <p>Electronegativity of Na = 6    Electronegativity of Cl = 3<br/>Difference is 3 <input checked="" type="checkbox"/></p> <p>With a difference of greater than 2.1 <input checked="" type="checkbox"/> electron transfer will take place and the bond <input checked="" type="checkbox"/> is ionic. <input checked="" type="checkbox"/> (value of 2.1 does not need to be given for this mark)</p> <p>Elektronegatiwiteit van Na = 6    Elektronegatiwiteit van Cl = 3<br/>Verskil is 3 <input checked="" type="checkbox"/></p> <p>Met 'n verskil van meer as 2.1 <input checked="" type="checkbox"/> sal elektroneoordrag plaasvind en die binding is ionies. <input checked="" type="checkbox"/> (waarde van 2.1 hoef nie vir hierdie punt gegee te word nie)</p> <p>4 marks/punte</p> | LO2 |

# Physical Science 2014 Control Test

**Lingjun Ying**



**Physical Science 2014 Control Test:**

The Top Books of the Year Physical Science 2014 Control Test The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous compelling novels captivating the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the fascinating narratives that have charmed audiences this year. Physical Science 2014 Control Test : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Physical Science 2014 Control Test : Delia Owens "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a masterful and suspenseful novel that will keep you speculating until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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