

AP Chemistry: A Sample Formal Laboratory Report

This paper is designed to help you prepare a chemistry lab report. Keep it in your chemistry notebook. All chemistry lab reports must be written neatly and well organized to receive full credit. Lab reports may be written or typed. It is highly recommended that you use graph or engineering bond paper for written reports.

Laboratory #7: Quantitative Determination of an Empirical Formula

I. Hypothesis: If nitric acid is poured onto tin, a tin oxide will be produced. If we know the initial mass of the tin metal and the mass of the final product, we can determine the empirical formula of the tin oxide product. There should be a whole number ratio between oxygen and tin. *(The hypothesis explains what is to be tested and will be written after reading the entire laboratory worksheet.)*

II. Equipment: *(Non-chemical equipment used in the experiment.)*
evaporating dish
watch glass
stirring rod
burner with ring stand, ring and wire gauze
forceps
beaker
balance

III. Reagents: *(A listing of chemicals used in the experiment with their amounts and any warnings.)*
tin metal (granulated) ~2 g.
5 cm³ (mL) nitric acid (HNO₃) **caution! severe burns**

IV. Procedure: Each step of the procedure must be written here. You may paraphrase and shorten the explanations, but the reader must be able to perform the experiment from these instructions. *(The procedure must be read carefully before the lab begins. Drawings of the experimental setup may be included here. The teacher may make changes to the procedure; make sure that you write any changes down!)*

V. Data: *(If the laboratory report is handwritten use a ruler to draw data tables and graphs! Always include units with all data entries.)*

	Procedure	Trial 1	Trial 2
a	mass of dish, and watch glass	74.14 g	
b	mass of dish, glass and tin	76.20 g	
c	mass of tin =b-a	2.06 g	
d	moles of tin	.0173 mol	
e	mass of dish, glass, and product	76.76 g	
f	mass of oxygen =e-b	.56 g	
g	moles of oxygen	.0350 mol	
h	mole ratio	2.02 : 1	
i	accepted ratio	2 : 1	
j	% error	1.00 %	

Sample Chemistry Laboratory Report

Judd E. Hollander



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Addison-Wesley Chemistry Laboratory Manual Dennis D. Staley,1995 **Sampling and Analysis of Rain** S.A. Campbell,1981 **Combined Compilation of Meat and Poultry Inspection Issuances for ...** ,1991 Science Educator's Guide to Laboratory Assessment Rodney L. Doran,2002 The book opens with an up to date discussion of assessment theory research and uses Then comes a wealth of sample assessment activities in biology chemistry physics and Earth science Keyed to the National Science Education Standards the activities include reproducible task sheets and scoring rubrics All are ideal for helping students reflect on their own learning during science lab Embankment Dam Instrumentation Manual Charles L. Bartholomew,1987 **Report** United States. Congress. House,1939 Scientific and Technical Aerospace Reports ,1995-05 **Report** National Center for Radiological Health (U.S.). Interlaboratory Technical Advisory Committee,1965 NTP GMM. ,2005 *Compilation of Meat and Poultry Inspection Issuances* ,1985-11

Water-resources Investigations Report ,2003 Consolidated Annual Report on State and Territorial Public Health Laboratories ,1975 Code of Federal Regulations United States. Department of the Interior,2000 Special edition of the Federal register containing a codification of documents of general applicability and future effect as of with ancillaries

Semiannual Report to the Congress United States. Environmental Protection Agency. Office of the Inspector General,1990 Annual Reports of Officers, Boards, and Institutions of the Commonwealth of Virginia, for the Year Ending September 30 ... Virginia,1928 Manual for the Certification of Laboratories Analyzing Drinking Water ,1982 **The Basic Practice of Statistics** David S. Moore,2010 This is a clear and innovative overview of statistics which emphasises major ideas essential skills and real life data The organisation and design has been improved for the fifth edition coverage of engaging real world topics has been increased and content has been updated to appeal to today s trends and research

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consultant and expert witness in civil and criminal cases **For All Practical Purposes** Consortium for Mathematics and Its Applications (U.S.), 2003 The sixth edition of the acclaimed classroom favorite offer a number of new features to help instructors strengthen the mathematical literacy of their students

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