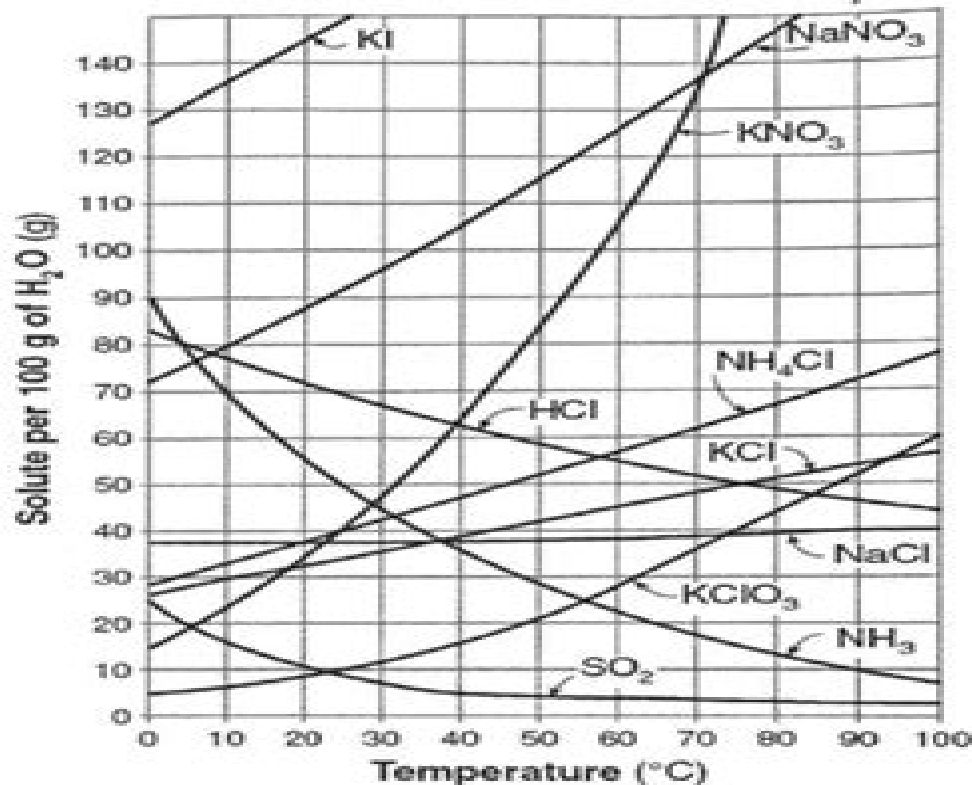


## Reading a Solubility Chart

- 1) The curve shows the # of grams of solute in a saturated solution containing 100 mL or 100 g of water at a certain temperature.
- 2) Any amount of solute below the line indicates the solution is unsaturated at a certain temperature
- 3) Any amount of solute above the line in which all of the solute has dissolved shows the solution is supersaturated.
- 4) If the amount of solute is above the line but has not all dissolved, the solution is saturated and the # grams of solute settled on the bottom of the container = total # g in solution - # g of a saturated solution at that temperature. (according to the curve)
- 5) Solutes whose curves move upward w/ increased temperature are typically solids b/c the solubility of solids increases w/ increased temperature.
- 6) Solutes whose curves move downward w/ increased temperature are typically gases b/c the solubility of gases decreases with increased temperature.



## Solubility Problems

1. Which of the salts shown on the graph is the most soluble in water at 10°C? **KI**
2. At what temperature is 90g of NaNO<sub>3</sub> a saturated solution in 100grams of water? **≈ 23°C**
3. A solution has 70grams of KNO<sub>3</sub> and is unsaturated, what is the minimum temperature? **46°C**
4. At 20°C, a saturated solution of KClO<sub>3</sub> is created in water. How many grams of potassium chlorate must be added to saturate the solution at 50°C? **12g**
5. Which of the salts shown on the graph is the least soluble in water at 10°C? **KClO<sub>3</sub>**
6. Which of the salts shown on the graph has the greatest increase in solubility as the temperature increases from 30 degrees to 60 degrees? **KNO<sub>3</sub>**

# Reading The Solubility Chart Problems Answers With Work

**H Kauffman**



## **Reading The Solubility Chart Problems Answers With Work:**

This book delves into Reading The Solubility Chart Problems Answers With Work. Reading The Solubility Chart Problems Answers With Work is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Reading The Solubility Chart Problems Answers With Work, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
    - Chapter 1: Introduction to Reading The Solubility Chart Problems Answers With Work
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    - Chapter 3: Reading The Solubility Chart Problems Answers With Work in Everyday Life
    - Chapter 4: Reading The Solubility Chart Problems Answers With Work in Specific Contexts
    - Chapter 5: Conclusion
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