

Sample paper - 1

GENERAL INSTRUCTION

1. You will not be allowed to write during the 15 minutes. This time is being spent in reading the question paper.
2. The time given at the head of this paper is the time allowed for writing the answer.
3. Attempt all questions from Section A and any 4 questions from Section B.
4. The intended marks for questions or parts of questions are given in brackets []

Time: 2Hrs
Marks:80

Max.

SECTION A [40 marks]

1.
 - (a) Fill in the blanks from the choices given in brackets.
 - (i) An element with atomic number 17 belongs to.....period.(first, third)
Sol:
Third
 - (ii) Write the electronic configuration of magnesium atom after losing 2 electrons.....(neon, argon)
Sol:
Neon
 - (iii) Whose example is ammonium hydroxide.....? (weak, strong)
Sol:
Weak
 - (iv) Write the oxidized product when the H_2S reacts with dil. HNO_3
(sulphur, nitrogen)
Sol:
Sulphur
 - (v) Write the molecular formula of the compound when ethyne reacts with chlorine ($C_2H_4Cl_2$, $C_2H_2Cl_2$)
Sol:
 $C_2H_2Cl_2$
 - (b) Choose the correct answer from the option
 - (i) Which law groups the elements in family of threem
(a) Law of traids

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SA Adler



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