

Using Special Right Triangles



Section A Use the triangle diagrams to help complete the table for the exact trig values.



	0°	30°	45°	60°	90°
sin	0				1
cos	1				0
tan	0				

Section B Give the exact values.

1) $\sin 30 + \cos 60$	2) $\tan 45 + \cos 60$	3) $4 \sin 30 + \cos 0$
4) $2 \sin 60$	5) $8 \cos 45$	6) $(\sin 45)^2 + 3 \cos 60$
7) $(3 \tan 30)^2 + (\tan 60)^2$	8) $5 \sin 60 - 3 \cos 30$	9) $9 \sin 30 - 3 \sin 90 + 4 (\cos 45)^2$

Section C Use the ratios of the given lengths to find the missing angles.

1) $a = \underline{\hspace{2cm}}$	2) $b = \underline{\hspace{2cm}}$	3) $c = \underline{\hspace{2cm}}$
4) $d = \underline{\hspace{2cm}}$	5) $e = \underline{\hspace{2cm}}$	6) $f = \underline{\hspace{2cm}}$

Practice B Spplying Special Right Triangles Answer Key

A Loxley



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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent a $i2o5$
 $5co\ i2\ 5co2\ i\ 2\ o\ 5\ 5\ c\ o\ i\ 2\ 5\ c\ o\ 2$

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons

chapter 20 worksheet redox beverly hills high school - Jul 02 2022

web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also $2sr\ o2\ 2sro\ sr0\ to\ sr2$ oxidized reducing agent $o0\ to\ o2$ reduced ox ag $2li\ s\ li2s\ li0\ to\ li1$ oxidized red ag $s0\ to\ s2$ reduced ox ag $3\ 2cs\ br2$