

Physical Science EOCT Study Guide – Domain 3

Physical Energy, Force, and Motion

Standards: PS.3-7 and 8

- i. Students will understand relationships and flow of energy within a system.
 - a. Explain thermal energy (heat) and describe the principle of conservation of energy as it applies to heat and temperature for two objects.
 - b. Explain heat. Thermal energy flows from objects from warmer to cooler objects.

described by $Q = mc\Delta T$ where Q is heat energy, m is mass, and c is specific heat.
 - c. Explain specific heat. The amount of heat energy needed to raise the temperature of 1 kg of substance by 1°C.
 - d. What is the formula for determining that change in heat that energy can hold?

$Q = mc\Delta T$ or $Q = L + c\Delta T$ $\Delta T = T_2 - T_1$
 - e. A sample of 0.5 kg of water at 20°C and specific heat of 4186 J/kg°C. Find the change in heat energy if the block is 2 mmol from 10°C to 20°C.

$Q = mc\Delta T$ $Q = (0.5)(4186)(20 - 10)$ $Q = 41,860 \text{ J}$
 - f. Explain conduction, convection, and radiation. Give examples of each.

Conduction is a transfer of heat through direct contact. Example: a metal spoon in a pot of soup.

Convection is heat transfer by movement of fluid. Example: a pot of boiling water.

Radiation is heat transfer by electromagnetic waves. Example: the sun's rays.

- ii. Explain the following phase change diagram.



- iii. Students will determine relationships among force, mass and motion.
 - a. Explain the difference between distance and displacement.

Distance is the total path length. Displacement is the change in position.
 - b. What is the difference between average speed and instantaneous speed?

Average speed is total distance divided by total time. Instantaneous speed is speed at a specific time.
 - c. What is the formula for average speed?

$v = \frac{d}{t}$

$$v = \frac{d}{t}$$

Physical Science Eoct Study Guide Energy Transformations

Matthew Yen



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