



MERCY JUNIOR COLLEGE  
Grade 11  
Physical Science  
Midterm Exam

NAME: \_\_\_\_\_  
YEAR LEVEL & SECTION: \_\_\_\_\_  
TEACHER'S NAME: \_\_\_\_\_  
PROCTOR'S NAME: \_\_\_\_\_

I. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

- It is a subatomic particle, symbol  $e^-$  or  $\beta^-$ , with a negative elementary electric charge.  
A. Neutron  
B. Electron  
C. Proton  
D. Atom
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.  
A. Neutron  
B. Electron  
C. Proton  
D. Atom
- It is a subatomic particle, symbol  $p$  or  $p^+$ , with a positive electric charge of  $+1e$  elementary charge and mass slightly less than that of a neutron.  
A. Neutron  
B. Electron  
C. Proton  
D. Atom
- It is a subatomic particle, symbol  $n$  or  $n^0$ , with no net electric charge and a mass slightly larger than that of a proton.  
A. Neutron  
B. Electron  
C. Proton  
D. Atom
- What type of nuclear reaction is this?



- Alpha Decay
- Beta Decay
- Gamma Ray
- No nuclear reaction

- It is a type of radioactive decay in which an atomic nucleus emits an alpha particle and thereby transforms or 'decays' into an atom with a mass number that is reduced by four and an atomic number that is reduced by two.  
A. Alpha Decay  
B. Beta Decay  
C. Gamma Ray  
D. Radiation

- It is a type of radioactive decay in which a beta ray (fast energetic electron or positron) and a neutrino are emitted from an atomic nucleus.

- Alpha Decay
- Beta Decay
- Gamma Ray
- Radiation

- What type of nuclear reaction is this?



- Alpha Decay
- Beta Decay
- Gamma Ray
- No nuclear reaction

- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.

- Electron Configuration
- Periodic Table of Elements
- Orbital Models
- Energy Levels

- What type of nuclear reaction is this?



- Alpha Decay
- Beta Decay
- Gamma Ray
- No nuclear reaction

# Physical Science 2june Exam Paper 1

**D Kirk**



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