

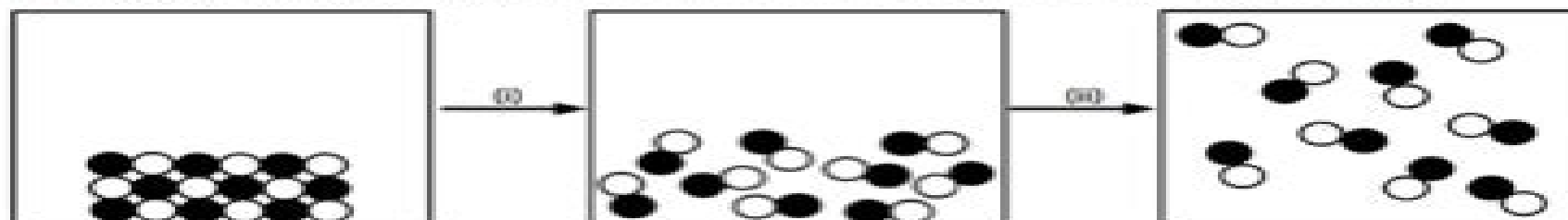
**QUESTION 1**

A number of answers are provided for each question.

Read the question and choose the most suitable answer from those provided.

Write only the letter that agrees with your answer next to the question number.

Questions 1.1 to 1.3 relate to a substance and its phase changes as seen in the diagrams below. THREE representations of particles of the same substance are provided for a physical change.



1.1 The substance in the three representations is a/an:

- A. element.
- B. mixture.
- C. compound.
- D. None of the above.

1.2 The processes occurring at points (i) and (ii) are:

	(i)	(ii)
A.	evaporation	condensation
B.	melting	evaporation
C.	solidification	condensation
D.	condensation	solidification

# Online Physical Science Worksheets

# Physical Science P2014 Grd1exampler

**Ensheng Dong**



**Physical Science P2014 Grd1exampler:**

## Unveiling the Power of Verbal Beauty: An Emotional Sojourn through **Physical Science P2014 Grd1exampl**

In a world inundated with displays and the cacophony of fast interaction, the profound energy and psychological resonance of verbal beauty frequently disappear into obscurity, eclipsed by the constant assault of noise and distractions. Yet, set within the lyrical pages of **Physical Science P2014 Grd1exampl**, a captivating work of fictional beauty that impulses with natural emotions, lies an unforgettable trip waiting to be embarked upon. Penned with a virtuoso wordsmith, this enchanting opus guides visitors on a psychological odyssey, gently exposing the latent potential and profound impact embedded within the elaborate web of language. Within the heart-wrenching expanse of the evocative examination, we will embark upon an introspective exploration of the book's key themes, dissect their captivating publishing fashion, and immerse ourselves in the indelible effect it leaves upon the depths of readers' souls.

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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