

# Periodic Table Basics

**Step 1:** Complete the square for each element by filling in the atomic number, name, & atomic mass.

**Step 2:** Determine the number of protons, neutrons, and electrons in an atom of each element.

**Step 3:** Identify if the element is a solid, liquid, or gas at room temperature.

**Step 4:** Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

**Step 5:** List at least three physical or chemical properties for each element.

**Step 6:** List at least three uses for each element.

**Step 7:** Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

**Step 8:** Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na  
Orange = B & Al

Pink = O & S  
Red = C & Si

Blue = Be & Mg  
Tan = N & P

Purple = F & Cl  
Yellow = He, Ne, & Ar

**Step 9:** Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

① <div style="text-align: center; font-size: 2em;">B</div>  	P= ② N= E=	③ O S O L O G
	M.P. = B.P. = ④	
Properties ⑤		
Uses ⑥		
Bohr Diagram      Lewis Structure		
<div style="display: flex; align-items: center; justify-content: space-around;">  <div style="text-align: center;">         ⑦   <div style="font-size: 1.5em;">B</div> </div> </div>		

Done? Answer the questions on the worksheet using the information on your Periodic Table!

# Physical Science If8767 Answers Percentage Composition

**L Reisser**



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## **Table of Contents Physical Science If8767 Answers Percentage Composition**

1. Understanding the eBook Physical Science If8767 Answers Percentage Composition
  - The Rise of Digital Reading Physical Science If8767 Answers Percentage Composition
  - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Answers Percentage Composition
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physical Science If8767 Answers Percentage Composition
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Answers Percentage Composition
  - Personalized Recommendations
  - Physical Science If8767 Answers Percentage Composition User Reviews and Ratings
  - Physical Science If8767 Answers Percentage Composition and Bestseller Lists
5. Accessing Physical Science If8767 Answers Percentage Composition Free and Paid eBooks
  - Physical Science If8767 Answers Percentage Composition Public Domain eBooks
  - Physical Science If8767 Answers Percentage Composition eBook Subscription Services
  - Physical Science If8767 Answers Percentage Composition Budget-Friendly Options

6. Navigating Physical Science If8767 Answers Percentage Composition eBook Formats
  - ePub, PDF, MOBI, and More
  - Physical Science If8767 Answers Percentage Composition Compatibility with Devices
  - Physical Science If8767 Answers Percentage Composition Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physical Science If8767 Answers Percentage Composition
  - Highlighting and Note-Taking Physical Science If8767 Answers Percentage Composition
  - Interactive Elements Physical Science If8767 Answers Percentage Composition
8. Staying Engaged with Physical Science If8767 Answers Percentage Composition
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physical Science If8767 Answers Percentage Composition
9. Balancing eBooks and Physical Books Physical Science If8767 Answers Percentage Composition
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physical Science If8767 Answers Percentage Composition
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Physical Science If8767 Answers Percentage Composition
  - Setting Reading Goals Physical Science If8767 Answers Percentage Composition
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science If8767 Answers Percentage Composition
  - Fact-Checking eBook Content of Physical Science If8767 Answers Percentage Composition
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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