

# Guided Reading and Study Workbook

**Differentiated Instruction**

**Solutions for All Learners**

## Science Explorer

# PRENTICE HALL Physical Science

### Guided Reading and Study Workbook

- Promotes active reading and enhances students' study skills using innovative questioning strategies and exercises
- Builds a record of students' work to use as a study aid for quizzes and tests
- Differentiates instruction through a wide range of question formats to reach a wide variety of learners
- Gives parents a handy resource to help students study and learn
- Answer Key located in the All-in-One Teaching Resources

# Physical Science Guided Reading And Study Workbook

## 15

**Yi-Tong Ma**



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web the balanced chemical equation for this reaction is  $\text{Mg}(\text{s}) + 2\text{HNO}_3(\text{aq}) \rightarrow \text{Mg}(\text{NO}_3)_2(\text{aq}) + \text{H}_2(\text{g})$  in this equation the s symbolizes solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the

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 b  $\text{mg}^{2+} \text{ aq} + \text{co}_3^{2-} \text{ aq} + 2\text{h}^+ \text{ aq} \rightarrow \text{mg}^{2+} \text{ aq} + \text{no}_3^- \text{ aq}$  mg write the balanced equation when nitric acid reacts with calcium hydroxide to produce water and calcium nitrate

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