

# Self Assessment Paper-3

Science

Class- X, Session: 2022-23

**UNSOLVED**

Time Allowed: 3 hours

Maximum Marks: 40

## General Instructions:

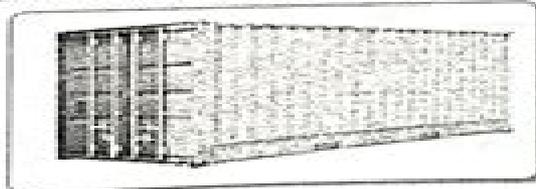
- This question paper consists of 39 questions in 5 sections.
- All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- Section A consists of 20 objective type questions carrying 1 mark each.
- Section B consists of 6 Very Short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- Section C consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- Section D consists of 3 Long Answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- Section E consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

### Section-A

20 Marks

(Select and write one most appropriate option out of the four options given for each of the questions 1 – 20)

- Three beakers labelled as A, B and C each containing 25 mL of water was taken. A small amount of NaOH, anhydrous  $\text{CuSO}_4$  and NaCl were added to the beakers A, B and C respectively. It was observed that there was an increase in the temperature of the solutions contained in beakers A and B, whereas in case of beaker C, the temperature of the solution falls. Which one of the following statement(s) is (are) correct? 1
  - In beakers A and B, exothermic process has occurred.
  - In beakers A and B, endothermic process has occurred.
  - In beaker C, exothermic process has occurred.
  - In beaker C, endothermic process has occurred.(A) (i) only (B) (ii) only  
(C) (i) and (iv) (D) (iii) and (iii)  
**OR**
- Ethane, with the molecular formula  $\text{C}_2\text{H}_6$  has: 1  
(A) 6 covalent bonds (B) 7 covalent bonds  
(C) 8 covalent bonds (D) 9 covalent bonds
- Shown below is a container that is used in the transportation of goods over long distances. 1



These containers are made of steel. Which property of steel is mainly used to make these containers?

- (A) Its ductility (B) Its malleability  
(C) Its metallic lustre (D) Its electrical conductivity
- In the given chemical reactions, Salt X  $\rightarrow$  Salt Y + Gas Z 1
  - Salt X is commonly used in bakery product.
  - Salt Y is common constituent of washing powder.
  - Gas Z turns lime water milky.

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**Yan Bai**



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