

TOEFL Independent Writing Sample

College is a place **where** students expand their knowledge and undergo new experiences. **Of course**, different people have different reasons for **studying** in college. For example, some people want to go on to post graduate studies, some hope to improve their job prospects, and some simply want to change their present situation. **In my opinion**, further education sets a person up for a better future, **regardless** of their motivation.

First of all, in terms of **gaining** knowledge, institutions of higher education expose students to a range of instructors, all of **whom** have **their** own area of expertise to share and with **which** **they** can help **their** students solve problems. **With** their help, students learn both basic and professional knowledge **which** may prove to be very useful **as** they go on to both work and contribute to society.

Secondly, in terms of new experiences, students learn how to prioritize and arrange their own time reasonably. Until college, **students'** lives are typically **organized** by their parents, **while** their studies are often arranged by their teachers. At university, students become solely responsible for decisions about both **living** and **studying**, **including** **deciding** **when** to get up, participate in sports, and complete assignments. **This** ability to prioritize, schedule, and perform is a critical life skill for future success.

Thirdly, **studying** in the college offers unique **socializing** opportunities. Everything from **sharing** accommodation to **completing** group-work demands a higher level of cooperation than most people experience elsewhere. **As** students acquire these invaluable communication skills, **they** become better prepared to contribute fully to the global society.

All in all, anyone **who** opts for further education enters a world of new experiences **that** will broaden their horizons, increase their knowledge base, and strengthen their communication skills; **thereby**, **setting** them up for successful futures.

Gerunds

Participles

Subordinating conj.

Conjunctive adverbs

References

Parallels

Subjunctives

Correlative Conj.

Possessives

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L Reisser



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negative when heat flows from the system to the surroundings thus exothermic reactions have a negative Δh of reaction the
word exothermic is derived from exo the greek word for outside and therme the greek word

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web in the context of a chemical reaction the term neutralization is used for a reaction between an acid and a base or alkali
historically this reaction was represented as acid base alkali salt water for example $\text{hcl naoh nacl h}_2\text{o}$

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web during this reaction the h^+ ions from the acid and the oh^- ions from the alkali react together to form water as this happens
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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable measure an appropriate volume of each liquid eg 25 cm³ 3 place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with HCl as 57.9 kJ mol⁻¹

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic eg 1 HCl(aq) + NaOH(aq) → NaCl(aq) + H₂O(l) for which ΔH = 57 kJ mol⁻¹ eg 2 HNO₃(aq) + KOH(aq) → KNO₃(aq) + H₂O(l) for which ΔH = 57 kJ mol⁻¹

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