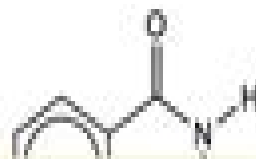


Which pair of statements about electrons in this molecule is correct?



In which group of ions and molecules are electrons delocalized in all species?

- A. CH_3COOH , O_2 , C_{60}
 B. CH_3COO^- , NO_2^- , $\text{C}(\text{graphite})$

Number of non-bonding pairs of electrons

- | | |
|----|---|
| A. | 3 |
| B. | 3 |
| C. | 5 |
| D. | 5 |

IB CHEM PAPER 1

PRACTICE

QUESTIONS

Which is the preferred Lewis for

- A. $\text{N}=\text{N}=\text{O}$
 B. $\text{:N}\equiv\text{N}-\text{O:}$
 C. $\text{:N}-\text{N}\equiv\text{O:}$
 D. $\text{N}=\text{O}=\text{N}$

ed between magnesium ions and

try of the central atom in SF_6Cl_2 ?

12. Which of the following is

- C. hexagonal
 D. octahedral

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