

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

G.12

**NATIONAL
SENIOR CERTIFICATE**

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reagent used in reaction IV. (1)



4.8 Draw the half-cell reaction for the zinc electrode. (1)



4.9 Identify the reaction type.

Write a balanced chemical equation for the reaction.

4.10 Draw a half-cell reaction for the copper electrode. (1)

4.11 The simplified diagram below represents an electrochemical cell. The electrodes are made of copper.

Physical Science Memo For Preliminary 2014

N Colangelo



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Table of Contents Physical Science Memo For Preliminary 2014

1. Understanding the eBook Physical Science Memo For Preliminary 2014
 - The Rise of Digital Reading Physical Science Memo For Preliminary 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Memo For Preliminary 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Memo For Preliminary 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Memo For Preliminary 2014
 - Personalized Recommendations
 - Physical Science Memo For Preliminary 2014 User Reviews and Ratings
 - Physical Science Memo For Preliminary 2014 and Bestseller Lists
5. Accessing Physical Science Memo For Preliminary 2014 Free and Paid eBooks

- Physical Science Memo For Preliminary 2014 Public Domain eBooks
- Physical Science Memo For Preliminary 2014 eBook Subscription Services
- Physical Science Memo For Preliminary 2014 Budget-Friendly Options
- 6. Navigating Physical Science Memo For Preliminary 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Memo For Preliminary 2014 Compatibility with Devices
 - Physical Science Memo For Preliminary 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Memo For Preliminary 2014
 - Highlighting and Note-Taking Physical Science Memo For Preliminary 2014
 - Interactive Elements Physical Science Memo For Preliminary 2014
- 8. Staying Engaged with Physical Science Memo For Preliminary 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Memo For Preliminary 2014
- 9. Balancing eBooks and Physical Books Physical Science Memo For Preliminary 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Memo For Preliminary 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Memo For Preliminary 2014
 - Setting Reading Goals Physical Science Memo For Preliminary 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Memo For Preliminary 2014
 - Fact-Checking eBook Content of Physical Science Memo For Preliminary 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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